# <span id="page-0-0"></span>Exploitation of in Situ Generated Sugar-Based Olefin Keto-Nitrones: Synthesis of Carbocycles, Heterocycles, and Nucleoside Derivatives

Soumendra Nath Das,† Arpan Chowdhury,‡ Neha Tripathi,‡ Prithwish K. Jana,† and Sukhendu B Mandal<sup>\*,†</sup>

† Chemistry Division, CSIR-Ind[ian](#page-10-0) Institute of Chemical Biology, 4, Raja S. C. Mullick Road, Jadavpur, Kolkata 700 032, India ‡ National Institute of Pharmaceutical Education and Research, 4, Raja S. C. Mullick Road, Jadavpur, Kolkata 700 032, India

**S** Supporting Information

[AB](#page-10-0)STRACT: [Application o](#page-10-0)f intramolecular 1,3-dipolar nitrone cycloaddition reaction on carbohydrate-derived precursors containing an olefin functionality at C-1 or C-3 or C-5 and a nitrone moiety at C-2 or C-3 as appropriate has resulted in the formation of structurally new cycloaddition products containing furanose-fused oxepane, thiepane, azepane, cyclopentane, cycloheptane, tetrahydrofuran, and pyranose-fused tetrahydrofuran rings. The structure and stereochemistry of these products have been characterized by spectral as well as single-crystal X-ray analyses. Two of the compounds have been transformed to the bicyclic nucleoside derivatives applying Vorbrü ggen reaction conditions.

# **NHACOAC** ۵Δ٢ **ÑHAc**  $ACOX = O$

# **ENTRODUCTION**

Both the inter- and intramolecular nitrone cycloaddition (INC) reactions provide efficient methods for the synthesis of natural as well as unnatural molecules.<sup>1</sup> However, the method that uses the chirality of natural products as chiral pool generates enantiomerically pure mol[ec](#page-10-0)ules of varying nature and structures. $^{2}$  In this perspective, the supremacy of carbohydrates as a chiral pool, is well-established. Various potential chiral auxiliaries, $3$  used for the synthesis of chiral molecules, have been generated from carbohydrates through judicious manipulation of the su[ga](#page-11-0)r backbone. Attempts to elaborate the synthetic utility of the INC methodology on sugar-based substrates have resulted in generating carbocycles,<sup>4</sup> heterocycles,<sup>5</sup> bicycles,<sup>6</sup> spirocycles,<sup>7</sup> cyclic ethers,<sup>8</sup> alkaloids,<sup>9</sup> amino acids,<sup>10</sup> nucleo $sides, <sup>11</sup>$  iminosugars,<sup>12</sup> pseudosac[ch](#page-11-0)arides,<sup>13</sup> enz[ym](#page-11-0)e inhi[b](#page-11-0)itors,<sup>14</sup> pre[cu](#page-11-0)rsors for pro[st](#page-11-0)aglandin<sup>1[5](#page-11-0)</sup> and tetrodot[oxi](#page-11-0)n,<sup>16</sup> and other [r](#page-11-0)elated molec[ular](#page-11-0) entities. Neverthe[les](#page-11-0)s, the chemistry utili[zin](#page-11-0)g this method continues [un](#page-11-0)abated in const[ruc](#page-11-0)ting complex ring systems. Synthetic applications of aldo-nitrones, generated from sugars, have been prevalent, although utilization of the reaction on sugar-based keto-nitrones<sup>18</sup> remai[ns](#page-11-0) relatively unexplored.

We envisage a strategy that involves an olefin [moi](#page-11-0)ety (allyl, homoallyl, O-allyl, S-allyl, N-allyl) at C-5 and a nitrone unit at C-3 (Path A), or an O-allyl group at C-1 and a nitrone moiety at C-2 (Path B), or an O-allyl group at C-3 and a nitrone function at C-2 (Path C) of sugar-derived substrates that could undergo INC reaction (Figure 1) to generate a diverse nature of carbocycles and heterocycles including tetrahydro-furan (-pyran) rings. The latter two subunits<sup>19</sup> are extensively found



Figure 1. A general strategy for construction of rings using INC reaction.

in a large number of bioactive natural products, such as  $C_{13}$ polyketides,<sup>20</sup> mono and diterpenoids,<sup>21</sup> lignans,<sup>22</sup> and ezomycin octosyl nucleoside, $23$  including several synthetic potent HIV[-1](#page-11-0) protease inhibitors,<sup>24</sup> such [as](#page-11-0) darunavi[r,](#page-11-0) UIC-94003, and GRL-0476, while [an](#page-11-0) oxepane ring is present in

Received: November 24, 2014 Published: December 16, 2014



a<br>Regents and conditions: (a) allyl bromide, DCM: 50% aq. NaOH (1:1), TBAB, rt, overnight; (b) DDQ, DCM:H20 (20:1), rt, 2 h; (c) DMP, DCM, 0 °C, 2 h; (d) BnNHOH, toluene, reflux, 10 h.





a<br>Reagents and conditions: (a) KSAc, DMF, rt, 3 h; (b) NaBH<sub>4</sub>, allyl bromide, NaOMe, dry MeOH, 0 °C−rt, 3 h; (c) DDQ, DCM:H<sub>2</sub>0 (20:1), rt, 2 h; (d) oxalyl chloride,  $CH_2Cl_2$ , −65 °C, DMSO, 2 h; (e) BnNHOH, toluene, reflux, 10 h.

some biologically important natural molecules, $25$  such as heliannuol B and C, sodwanone S, and zoapatanol. $^{26}$  Many sulfur heterocycles<sup>27</sup> as well used as drugs display b[iol](#page-11-0)ogical and synthetic importances.<sup>28</sup> We report herein the results, [ob](#page-11-0)tained from exploration [of](#page-11-0) the INC reaction of carbohydrate-derived keto-nitrones via th[e s](#page-11-0)trategy as depicted, leading to the formation of new compounds, which contain isoxazolidinefused oxepane, azepane, thiepane, perhydrofurofuran, and carbocyclic rings of varied sizes. Two of these products have been converted to the bicyclic nucleoside analogues.<sup>29</sup>

### ■ RESULTS AND DISCUSSION

Prior to the application of olefin-keto-nitrone cycloaddition reactions on the backbone of carbohydrate derived substrates, preparation of some appropriate precursors from D-glucose was essentially necessary. The success of the strategy solely depended on the appendage of a heteroallyl or allyl moiety, and a hydroxyl group at the proper carbon of the substrates, prepared from  $1,2:5,6$ -di-O-isopropylidene- $\alpha$ -D-glucofuranose 1, through simple transformations. The hydroxyl group of the precursor was then oxidized to a ketone, followed by nitrone formation and cycloaddition reaction, furnishing the desired INC product.

Synthesis of Oxygen Heterocycle (via Path A). Allylation of  $2^{30}$  (prepared from 1 in three steps, viz. pmethoxybenzylation of the hydroxyl group, selective removal of the 5,6-O-isopr[op](#page-11-0)ylidene protection by acid treatment, vicinal diol cleavage, and reduction of the aldehyde moiety) with allyl bromide in the presence of 50% aqueous sodium hydroxide and tetrabutylammonium bromide (TBAB) as phase transfer catalyst in DCM solvent at room temperature furnished 3 in 96% yield (Scheme 1). Removal of the PMB protection $31$  by DDQ in a DCM−H2O mixture afforded 4 (78%). The hydroxyl group of 4 was oxidized<sup>32</sup> by Dess–Martin periodinane ([DM](#page-12-0)P)

in DCM to the corresponding ketone, which, without purification, was treated with N-benzyl hydroxylamine (BnNHOH) in refluxing toluene to furnish the isoxazolidinoxepane derivative 6 (55%) through the nonisolable intermediate 5. There could be two possible ways for the cyclization reaction (Figure 1). The observation of a peak at  $m/z$  356 (M + Na)<sup>+</sup> in the ESI mass spectrum confirmed the molecular weight of 6. The upfield proton signals at  $\delta$  2.24 and 2.74, and a carbon signal [at](#page-0-0)  $\delta$  31.6 in the <sup>1</sup>H and <sup>13</sup>C NMR spectra clearly indicated the cyclization involving an attack by the oxyanion at the methine terminal of the olefin moiety occurred. The cycloaddition via the other mode of cyclization furnishing the pyran derivative was discarded after analyses of the NMR spectra. The structure as well as the stereochemistry of 6 was further confirmed by the single-crystal X-ray analysis (Figure S1, Supporting Information).

Synthesis of Sulfur Heterocycle (via Path A). The iodide group of 7, which was prepared from 2 by treatment wit[h](#page-10-0)  $I_2/PPh_3/$  $I_2/PPh_3/$  $I_2/PPh_3/$  $I_2/PPh_3/$  imidazole,<sup>33</sup> was substituted by a thioacetyl group<sup>11c</sup> using KSAc in DMF to produce  $8$  in 91% yield (Scheme 2). Deprotect[ion](#page-12-0) of the acetate and subsequent allylat[ion](#page-11-0) of the thiolate anion in a one-pot reaction using  $NaBH<sub>4</sub>/allyl$  bromide/NaOCH<sub>3</sub> in dry MeOH furnished the thioallyl derivative 9 in 72% yield. Sodium borohydride maintained a reductive condition within the reaction mixture in preventing the formation of a disulfide bond. Removal of the PMB protection by DDQ produced 10 (80%). Various attempts including DMP to oxidize the hydroxyl group were unsuccessful due to the oxidation of sulfur. However, Swern oxidation<sup>34</sup> using oxalyl chloride/DMSO/Et<sub>3</sub>N in DCM afforded its corresponding ketone, which, without purification, was treat[ed](#page-12-0) with BnNHOH in refluxing toluene to produce the tetracyclic thiepane product 11 in 42% yield. The stereochemistry of the bridge methylene was assigned on the basis of the analogy with the corresponding oxepane derivative 6, obtained by cycloaddition reaction of the corresponding 5-Oallyl nitrone.

Synthesis of Nitrogen Heterocycle (via Path A). In a further manipulation of the strategy, the mesyl functionality of 12 (derived from its corresponding alcohol 2) was substituted by an azido group by heating with  $\text{NaN}_3$  in DMF, furnishing 13 in 94% yield (Scheme 3). Selective reduction of the azido group

#### Scheme 3. Construction of Azepane Ring in 18 on Sugar Backbone<sup>a</sup>



<sup>a</sup>Reagents and conditions: (a) NaN<sub>3</sub>, DMF, 80–100 °C, 6 h; (b) PPh<sub>3</sub>, moist-THF, reflux, 6 h; (c)  $(Boc)_2O$ , CH<sub>2</sub>Cl<sub>2</sub>, rt, 4 h; (d) allyl bromide, NaH, DMF, TBAB, rt, overnight; (e) DDQ, DCM:H2O (20:1), rt, 2 h; (f) DMP, DCM, 0  $^{\circ}$ C, 2 h; (g) BnNHOH, toluene, reflux, 12 h.

by Staudinger reaction<sup>35</sup> in moist THF with  $PPh_3$  furnished in 71% yield the primary amine 14, which was subsequently protected by di-tert-b[uty](#page-12-0)l dicarbonate to give 15 (81%). Nallylation of 15 using allyl bromide/NaH/TBAB in dry DMF furnished 16 in 88% yield. Deprotection of the PMB group by DDQ furnished the required alcohol precursor 17 (82%). DMP oxidation of the secondary hydroxyl group produced its corresponding ketone, which, upon nitrone formation with BnNHOH in refluxing toluene and subsequent cyclization, furnished the seven-membered tetracyclic azepane heterocycle 18 in 64% yield. All the products showed appropriate NMR spectral data. The stereochemistry of 18 was confirmed indirectly by single-crystal X-ray analysis of one of its derived products 46 (described in Scheme 12).

Synthesis of Carbocycles (via Path A). In a similar approach toward the synthesis of [a](#page-4-0) seven-membered carbocycle, the alkene  $19^{36}$  was readily oxidized to the alcohol 20  $(80%)$  by 9-BBN $^{37}$  in THF (Scheme 4). The hydroxyl group was oxidized by D[MP](#page-12-0) to obtain its corresponding aldehyde, which was readil[y c](#page-12-0)onverted to a mixture of the alcohols 21  $(1:1 \text{ ratio})$  in 77% yield in two steps using Barbier allylation<sup>38</sup> with allyl bromide in the presence of Zn dust in THF−NH4Cl. Acetylation of the mixture and removal of the PMB protecti[on](#page-12-0) by DDQ afforded 22, a mixture of isomeric products (∼1:1). Oxidation of the hydroxyl group of the mixture of compounds by DMP afforded their corresponding ketones, which were treated with BnNHOH in refluxing toluene to furnish the seven-membered carbocyclic derivative 23 in 40% overall yield. It was interesting to note that only one isomer underwent INC reaction, furnishing only one product. The X-ray analysis of 23 (Figure S2, Supporting Information) confirmed the stereochemistry of the bridge methylene as well as the acetoxy group.



a<br>Reagents and conditions: (a) 9-BBN, THF, 0 °C−rt, overnight,  $\text{H}_2\text{O}_2$ , NaOH; (b) DMP, DCM, 0 °C, 2 h; (c) allyl bromide, Zn-dust, THF:NH<sub>4</sub>CI (1:5), 0 °C−rt, 12 h; (d) Ac<sub>2</sub>0, DMAP, py, rt, 8 h; (e) DDQ, DCM:H20 (20:1), rt, 2 h; (f) BnNHOH, toluene, reflux, 10 h.

The successful formation of the seven-membered carbocycle ring prompted us to replace the homoallyl group by an allyl chain in order to obtain a six-membered carbocyclic ring. To this end, vicinal diol cleavage of 24, followed by Barberier allylation, hydroxyl group protection, and subsequent PMB deprotection (Scheme 5), furnished 25 (46% overall) as the

Scheme 5. Construction of Five-Membered Carbocycle Ring of  $27^a$ 



<sup>a</sup>Reagents and conditions: (a) NalO<sub>4</sub>, MeOH, 0 °C,1 h; (b) allyl bromide, Zn-dust, THF:NH4CI (1:5), 0 °C−rt, 12 h; (c) BnBr, NaH, DMF, TBAB, 0 °C−rt, 6 h; (d) DDQ, DCM:H<sub>2</sub>0 (20:1), rt, 2 h; (e) DMP, DCM, 0 °C, 2 h; (f) BnNHOH, toluene, reflux, 12 h.

single isomer. The reason for obtaining the only one isomer was due to the approach of the incoming allyl nucleophile from the least hindered  $\alpha$ -side. The  $\beta$ -side was hindered by pmethoxybenzyl protection at the C-3 position of the sugar ring as well as by blockage of this face due to the formation of a cyclic five-membered transition state<sup>39</sup> through the coordination of unshared electron pairs of oxygen (both carbonyl and ring oxygen) and allylzinc bromide. [Ox](#page-12-0)idation of the hydroxyl group by DMP and subsequent treatment with BnNHOH in refluxing toluene formed the nonisolable nitrone intermediate 26, which subsequently produced the cyclopentyl tetrahydrofuran derivative 27 (instead of six-membered carbocyclic ring) in 66% yield in two steps. The structure and stereochemistry of the product were confirmed by single-crystal X-ray analysis (Figure S3, Supporting Information). It is important to note that the perhydrocyclopentanofuran skeleton of 27 is present in an HIV pro[tease inhibitor GRL-0657](#page-10-0)9. $^{24}$ 

Construction of Perhydrofurofuran Ring (via Path B). For the construction of a hexahydrof[uro](#page-11-0) $[2,3-b]$ furan ring, the O-allylation at the anomeric center of  $28^{40}$  using allyl alcohol/

tosic acid afforded 29 (Scheme 6) as the single  $\alpha$ -isomer in 70% yield. The  $S_N$ 2 attack by the allyl alcohol from the β-face was

Scheme 6. Construction of Hexahydrofuro[2,3-b]furan Ring of  $30<sup>a</sup>$ 

$$
BnO \underbrace{\text{BnO}}_{BnO \text{ 28}} \underbrace{\text{BnO}}_{70\%} \underbrace{\text{BnO}}_{T0\%} \underbrace{\text{BnO}}_{BnO \text{ 29}} \underbrace{\text{O}}_{79\%} \underbrace{\text{BnO}}_{T9\%} \underbrace{\text{BnO}}_{BnO \text{ N}} \underbrace{\text{O}}_{30} \underbrace{\text{O}}_{30}
$$

<sup>a</sup>Reagents and conditions: (a) allyl alcohol,TsOH·H<sub>2</sub>O, reflux, 6 h; (b) DMP, DCM, 0 °C, 2 h; (c) BnNHOH, toluene, reflux,12 h.

hindered by the steric hindrance of the two bulky benzyl groups at C-3 and C-5, and therefore, the attack took place from the opposite  $α$ -face. Oxidation of the hydroxyl group by DMP, and subsequent nitrone generation by reaction with BnNHOH and cyclization, afforded the INC product 30 (79% in two steps). The structure and stereochemistry of 30 were confirmed by single-crystal X-ray analysis (Figure S4, Supporting Information).

In a comparable fashion for the synthesi[s of a hexahydrofuro-](#page-10-0) [\[3,4-](#page-10-0)b]furan ring (via Path C), the 1-deoxy sugar derivative  $31<sup>41</sup>$  upon oxidation of the hydroxyl group by DMP, followed by nitrone generation using BnNHOH and its in situ cy[cli](#page-12-0)zation, smoothly afforded the isoxazolidine-fused bisfuran derivative 32 in 62% yield (Scheme 7). The product was

Scheme 7. Construction of Hexahydrofuro  $[3,4-b]$  furan Ring of  $32^a$ 



<sup>a</sup>Reagents and conditions: (a) DMP, DCM, 0 °C, 2 h; (b) BnNHOH, toluene, reflux, 6 h.

characterized by NMR spectral analyses, and its stereochemistry at the ring juncture was determined by single-crystal X-ray analysis of one of its derived products 42.

Synthesis of Perhydropyranofuran Ring (via Path B). Based on the success in creating a furanofuran ring by INC reaction, an application of the strategy to construct a pyranofuran ring from the sugar-derived precursor was attempted. Thus, D-ribopyranose (33) was subjected to anomeric O-allylation using p-TSA in allyl alcohol, followed by acetonide formation by reaction with 2,2-dimethoxypropane/conc.  $H_2SO_4/Ag_2CO_3$  in acetone, affording 34 ( $\beta$ isomer) as the major isomer in 38% yield (Scheme 8). The

#### Scheme 8. Construction of Perhydropyranofuran Ring of  $35<sup>a</sup>$



<sup>a</sup>Reagents and conditions: (a) allyl alcohol, TsOH, reflux, 4 h; (b) 2,2dimethoxypropane, acetone, cone.  $H_2SO_4$ , AgCO<sub>3</sub>, 12 h; (c) DMP, DCM, 0 °C, 3 h; (d) BnNHOH, toluene, reflux, 12 h.

hydroxyl group was oxidized by DMP reagent to its corresponding ketone, which, upon reaction with BnNHOH, yielded the isoxazolidine-fused pyranofuran derivative 35 in 55% yield in two steps. The structure as well as the stereochemistry of the product was confirmed by single-crystal X-ray analysis (Figure S5, Supporting Information).

Cleavage of Isoxazolidine Rings of the INC Products. Installation of nucleoside [bases at the anomeric c](#page-10-0)enter of the INC products for the synthesis of bicyclic nucleoside derivatives required cleavages of the isoxazolidine rings and removal of the acetonide protection. Thus, treatment of the INC products 6, 18, and 27 with molybdenum hexacarbonyl  $(Mo(CO)_6)$  in refluxing aqueous MeCN<sup>42</sup> cleaved the isoxazolidine rings and removed the N-Bn protection to produce the corresponding amino alcohols [36](#page-12-0)−38 in ∼70− 80% yield (Scheme 9). However, the O-Bn group of 27 was not

# Scheme 9. Cleavage of the N−O Bond and Benzyl Protection from 6, 11, 18, and  $27<sup>a</sup>$



<sup>a</sup>Reagents and conditions: (a)  $Mo(CO)_{6}$  MeCN: H<sub>2</sub>O (15:1), reflux,12 h.

cleaved by the catalyst and the product 38 contained the O-Bn protection. On the other hand, only the isoxazolidine ring cleavage occurred in 11, furnishing 39 having an NHBn group at quaternary C-3 of the sugar moiety. Therefore, the hydrogenolytic method of cleavage using hydrogen gas over Pd/C was tried to remove the isoxazolidine ring as well as the benzyl protection of the INC products in one-pot.

Cleavage of the isoxazolidine rings and deprotection of the benzyl groups of 30 and 32 using hydrogen gas over Pd/C were successfully completed via catalytic hydrogenation reaction over Pd/C (10%) in MeOH to the corresponding amino alcohols, which, upon acetylation with  $Ac_2O/Py/DMAP$ , produced their corresponding bisfuran derivatives 40 and 41 in 74% and 71% yields (Scheme 10). However, deprotection of the N-Bn of 39 by hydrogenation reaction was unsuccessful due to sulfur poisoning of th[e ca](#page-4-0)talyst.

At this stage, it was felt that the free amino alcohol could be trapped by an aldehyde to a crystalline cyclized product to confirm the stereochemistry of the INC product 32, shown in Scheme 7. Thus, the dihydroxy amino alcohol 42, obtained after hydrogenation reaction upon 32 in MeOH, was treated with an aqueous solution of acetaldehyde to isolate 43 as a crystalline solid in 57% yield (Scheme 11). The doublet signal

<span id="page-4-0"></span>

<sup>a</sup>Reagents and condition: (a)  $10\%$  Pd/C, H<sub>2</sub>, MeOH, rt, 12 h; (b)  $Ac<sub>2</sub>0$ , pyridine, DMAP, rt, 6 h.

Scheme 11. Reaction of Dihydroxy Amino Alcohol 42 with Acetaldehyde<sup>a</sup>



<sup>a</sup>Reagents and conditions: (a)  $10\%$  Pd/C,  $H_2$ , MeOH, rt, 12 h; (b) Ac<sub>2</sub>O, pyridine, DMAP, rt, 6 h; (c) 40 wt % aq. CH<sub>3</sub>CHO.

 $(\delta$  1.31) for the methyl protons in the <sup>1</sup>H NMR spectrum confirmed the insertion of the acetaldehyde residue. The structure and stereochemistry of the product were confirmed by the single-crystal X-ray analysis (Figure S6, Supporting Information). The adduct 43, upon treatment with  $Ac_2O/Py/$ DMAP, furnished the triacetate bisfuran derivative 41 [in 91%](#page-10-0) yield.

[Synthesi](#page-10-0)s of Bicyclic Nucleoside Analogues 44 and 45. Toward the target, deprotection of the 1,2-acetonide moiety from 36 and 38 by acid treatment, followed by peracetylation in a one-pot reaction using  $Ac_2O-TfOH$ , afforded their respective acetylated products (anomeric mixture). The mixture, without further purification, was used for installation of a uracil base at the anomeric center via Vorbrüggen glycosidation reaction<sup>43</sup> (uracil, BSA, TMSOTf, MeCN, 50 °C) to furnish their corresponding bicyclic nucleoside derivatives 44 and 45 ([Sc](#page-12-0)heme 12).

However, an attempt to introduce a nucleoside base at the anomeric carbon of 46, derived from 37 by acetylation, failed to produce any desired nucleoside; instead, an intractable mixture of products was obtained. The structure of 46 was confirmed by single-crystal X-ray analysis, which also confirmed the structure of the INC product 18 (Figure S7, Supporting Information). Similarly, installation of the nucleoside base on 11 and 23 using Vorbrü ggen reaction through var[ious manipulations wa](#page-10-0)s unsuccessful.

#### ■ CONCLUSIONS

The work presented herein describes a potential application of INC reaction for the stereoselective synthesis of chiral heterocycles and carbocycles of varied nature using ketonitrone-olefins, which have been derived from D-glucose-based substrates. The structure and stereochemistry of the INC

Scheme 12. Synthesis of Bicyclic Nucleosides Derived from INC Products $a$ 



<sup>a</sup>Reagents and conditions: (a) Ac<sub>2</sub>O, TfOH, AcOH, 0 °C−rt, 2 h; (b) uracil, BSA, TMSOTf, MeCN, 50 °C, 17 h; (c) Ac<sub>2</sub>O, pyridine, DMAP, rt, 12 h.

products were confirmed by spectral and single-crystal X-ray analyses. Two of these products have been translated to the corresponding bicyclic nucleoside derivatives using Vorbrü ggen glycosidation reaction. However, several problems have occurred during nucleosidation reactions on some of the INC products. The ease of preparation of sugar-based precursors for INC reaction makes the method practical, efficient, and useful. The strategy seems valuable for the synthesis of other ring systems through judicious manipulation of the substrates.

# **EXPERIMENTAL SECTION**

General. Melting points were taken in open capillaries and are uncorrected.  ${}^{1}H$  and  ${}^{13}C$  NMR spectra were recorded in CDCl<sub>3</sub> and CD3OD as solvents using TMS as internal standard. Mass spectra were recorded using EI and ESI mode. Specific rotations were measured at 589 nm. Precoated plates (0.25 mm, silica gel 60  $F_{254}$ ) were used for thin-layer chromatography. Column chromatography was performed on silica gel (60−120, 100−200, and 230−400 mesh). All the solvents were distilled and purified as necessary.

(3aR,5R,6S,6aR)-5-(Allyloxymethyl)-6-(4-methoxybenzyloxy)-2,2-dimethyltetrahydrofuro[2,3-d][1,3]dioxole (3). TBAB  $(1.04 \text{ g}, 3.23 \text{ mmol})$  was added to a solution of  $2 (10.0 \text{ g}, 32.26 \text{ mmol})$ in DCM (80 mL), and the mixture was stirred at room temperature with portionwise addition of 50% aqueous NaOH solution (80 mL). Allyl bromide (3.35 mL, 38.71 mmol) was added to the mixture, and the resulting solution was stirred overnight at room temperature. The organic layer was separated from NaOH solution, washed with brine  $(3 \times 40 \text{ mL})$ , dried  $(Na_2SO_4)$ , and concentrated to an oil, which was purified by column chromatography on silica gel (60−120 mesh) using a mixture of petroleum ether−EtOAc (9:1) as eluent to furnish 3 (10.8 g, 96%) as a colorless liquid.  $[\alpha]_{D}^{25} - 38$  (c 0.39, CHCl<sub>3</sub>); <sup>1</sup>H NMR (CDCl<sub>3</sub>, 300 MHz):  $\delta$  1.32 (s, 3H), 1.48 (s, 3H), 3.68 (d, 2H, J = 6.0 Hz), 3.81 (s, 3H), 3.95 (d, 1H, J = 2.7 Hz), 3.97−4.09 (m, 2H), 4.36  $(m, 1H)$ , 4.45 (d, 1H,  $J = 11.7$  Hz), 4.58 (partially merged d, 1H,  $J =$ 3.3 Hz), 4.61 (d, 1H,  $J = 11.4$  Hz), 5.18 (d, 1H,  $J = 10.2$  Hz), 5.28 (dd, 1H, J = 0.9, 16.7 Hz), 5.84−5.95 (m, 2H), 6.68 (d, 2H, J = 8.4 Hz),

7.24 (d, 2H, J = 11.1 Hz); <sup>13</sup>C NMR (CDCl<sub>3</sub>, 75 MHz):  $\delta$  26.2 (CH<sub>3</sub>), 26.7 (CH<sub>3</sub>), 55.2 (CH<sub>3</sub>), 67.5 (CH<sub>2</sub>), 71.6 (CH<sub>2</sub>), 72.3 (CH<sub>2</sub>), 79.1 (CH), 81.2 (CH), 82.3 (CH), 105.0 (CH), 111.5 (C), 113.8 (2 × CH), 117.1 (CH<sub>2</sub>), 129.3 (2  $\times$  CH), 129.5 (C), 134.5 (CH), 159.3 (C); HRMS (ESI–QToF, positive ion) calcd for C<sub>19</sub>H<sub>26</sub>NaO<sub>6</sub>, m/z 373.1627, found 373.1656.

(3aR,5R,6S,6aR)-5-(Allyloxymethyl)-2,2-dimethyltetrahydro**furo[2,3-d][1,3]dioxol-6-ol (4).** DDQ (6.81 g, 30 mmol) was added to a solution of 3 (7.0 g, 20 mmol) in a mixture of  $DCM-H<sub>2</sub>O$  (20:1, 42 mL), and the solution was stirred at room temperature for 2 h. After quenching the reaction with a saturated  $NAHCO<sub>3</sub>$  solution (50 mL), the mixture was extracted with DCM  $(2 \times 50 \text{ mL})$ . The combined extract was dried  $(Na_2SO_4)$  and concentrated to a residue, which was purified chromatographically on silica gel (100−200 mesh) using petroleum ether−EtOAc (5:1) as eluent to give 4 (3.6 g, 78%) as a colorless oil.  $[\alpha]_{\text{D}}^{25} - 1$  (c 0.44, CHCl<sub>3</sub>); <sup>1</sup>H NMR (CDCl<sub>3</sub>, 300 MHz):  $\delta$  1.32 (s, 3H), 1.49 (s, 3H), 3.77 (d, 1H, J = 2.7 Hz), 3.86– 3.98 (m, 2H), 4.04 (dd, 1H,  $J = 6.0$ , 12.6 Hz), 4.12 (dd, 1H,  $J = 5.4$ , 12.9 Hz), 4.24 (brd, 1H,  $J = 2.7$  Hz), 4.30 (brs, 1H), 4.53 (d, 1H,  $J =$ 3.3 Hz), 5.23 (d, 1H,  $J = 10.8$  Hz), 5.29 (d, 1H,  $J = 17.7$  Hz), 5.83– 5.94 (m, 1H), 5.99 (d, 1H,  $J = 3.6$  Hz); <sup>13</sup>C NMR (CDCl<sub>3</sub>, 75 MHz):  $\delta$  26.0 (CH<sub>3</sub>), 26.6 (CH<sub>3</sub>), 68.0 (CH<sub>2</sub>), 72.8 (CH<sub>2</sub>), 76.1 (CH), 781 (CH), 85.2 (CH), 104.7 (CH), 111.5 (C), 117.9 (CH<sub>2</sub>), 133.6 (CH); HRMS (ESI-QToF, positive ion) calcd for  $C_{11}H_{18}NaO_5$ ,  $m/z$ 253.1052, found 253.1072.

(3S,6aS,7aR,10aR,10bR)-1-Benzyl-9,9-dimethylhexahydro-1H-3,10b-methano[1,3]dioxolo[4′,5′:4,5]furo[3,2-c][1,6,2] **dioxazocine (6).** DMP (6.9 g, 16.3 mmol) was added to a solution of 4 (2.5 g, 10.87 mmol) in DCM (20 mL) at 0  $\degree$ C under N<sub>2</sub>, and the solution was stirred for 2 h. The solvent was evaporated, and the residue was extracted with DCM  $(2 \times 40 \text{ mL})$ . The extract was washed with a saturated solution of  $NaHCO<sub>3</sub>$  (30 mL) and 10%  $Na<sub>2</sub>S<sub>2</sub>O<sub>3</sub>$  solution (30 mL), dried (Na<sub>2</sub>SO<sub>4</sub>), and evaporated to furnish a crude ketone. To a solution of this ketone in toluene (60 mL) was added BnNHOH (2.0 g, 16.31 mmol), and the mixture was heated at reflux for 10 h. The solvent was evaporated in rotary evaporator to a gummy residue, which was purified by column chromatography on silica gel (230−400 mesh). Elution with petroleum ether−EtOAc (17:3) furnished 6 (2.0 g, 55%) as a colorless solid. mp 192−193 °C;  $[\alpha]_D^{25}$  + 157 (c 0.37, CHCl<sub>3</sub>); <sup>1</sup>H NMR (CDCl<sub>3</sub>, 300 MHz):  $\delta$  1.36 (s, 3H), 1.60 (s, 3H), 2.24 (t, 1H,  $J = 10.5$  Hz), 2.74 (d, 1H,  $J = 11.7$  Hz), 3.60 (apparent t, 2H,  $J = 13.8$ , 17.1 Hz), 3.83 (apparent t, 2H,  $J = 14.4$ , 18.0 Hz), 4.20 (m, 3H), 4.60 (s, 1H), 4.70 (brd, 1H, J = 7.5 Hz), 5.94 (s, 1H), 7.30–7.48 (m, 5H); <sup>13</sup>C NMR (CDCl<sub>3</sub>, 75 MHz):  $\delta$  26.2  $(CH_3)$ , 26.5  $(CH_3)$ , 31.6  $(CH_2)$ , 57.5  $(CH_2)$ , 68.4  $(CH_2)$ , 72.6  $(CH_2)$ , 76.3 (C), 78.3 (CH), 79.6 (CH), 82.3 (CH), 103.9 (CH), 113.0 (C), 127.3 (CH), 128.4 (2  $\times$  CH), 129.1 (2  $\times$  CH), 137.7 (C); HRMS (EI, magnetic sector, positive ion) calcd for  $C_{18}H_{23}NO_5$ ,  $m/z$  333.1576, found 333.1575.

(3aR,5S,6R,6aR)-5-(Iodomethyl)-6-(4-methoxybenzyloxy)- 2,2-dimethyltetrahydrofuro[2,3-d][1,3]dioxole (7). To a solution of 2 (3.5 g, 11.3 mmol) in toluene (30 mL) were added  $Ph_3P$  (4.44 g, 16.95 mmol) and imidazole (1.54 g, 22.6 mmol), and the mixture was heated at 70 °C for 30 min. Iodine (4.3 g, 16.95 mmol) was added to it, and the heating was continued for 3 h. The mixture was cooled, and the solution was washed successively with 30%  $\text{Na}_2\text{S}_2\text{O}_3$  solution (20 mL) and water ( $3 \times 10$  mL), dried (Na<sub>2</sub>SO<sub>4</sub>), and evaporated to a crude residue, which was purified by column chromatography on silica gel (100−200 mesh). Elution was made with petroleum ether−EtOAc (9:1) to furnish 7 (4.2 mg, 88%) as a colorless oil.  $[\alpha]_D^{25} - 60$  (c 0.32, CHCl<sub>3</sub>); <sup>1</sup>H NMR (CDCl<sub>3</sub>, 300 MHz):  $\delta$  1.32 (s, 3H), 1.50 (s, 3H), 3.29 (s, 1H), 3.31 (d, 1H,  $J = 2.7$  Hz), 3.81 (s, 3H), 4.08 (d, 1H,  $J =$ 3.0 Hz), 4.43−4.53 (m, 2H), 4.61−4.63 (m, 2H), 5.95 (d, 1H, J = 3.6 Hz), 6.89 (d, 2H,  $J = 8.7$  Hz), 7.30 (d, 2H,  $J = 8.4$  Hz); <sup>13</sup>C NMR (CDCl<sub>3</sub>, 75 MHz):  $\delta$  – 0.96 (CH<sub>2</sub>), 26.2 (CH<sub>3</sub>), 26.8 (CH<sub>3</sub>), 55.2  $(CH<sub>3</sub>), 72.3$  (CH<sub>2</sub>), 81.0 (CH), 81.1 (CH), 81.9 (CH), 105.6 (CH), 111.8 (C), 113.8 (2 × CH), 129.2 (C), 129.6 (2 × CH), 159.4 (C); HRMS (ESI–QToF, positive ion) calcd for  $C_{16}H_{21}NaO_5$ ,  $m/z$ 443.0331, found 443.0333.

S-(((3aR,5S,6R,6aR)-6-(4-Methoxybenzyloxy)-2,2-dimethyltetrahydrofuro[2,3-d][1,3]dioxol-5-yl)methyl)ethanethioate (8). A solution of 7 (4.5 g, 10.7 mmol) in DMF (75 mL) was stirred at room temperature for 10 min, and then potassium thioacetate (3.66 g, 32.1 mmol) was slowly added to it over a period of 10 min. The mixture was stirred at room temperature for an additional 3 h. The solvent was evaporated to furnish a residue, which was extracted with CHCl<sub>3</sub> ( $3 \times 30$  mL). The combined extract was washed with brine ( $50$ ) mL), dried  $(Na_2SO_4)$ , and evaporated in vacuo. The crude product was purified by column chromatography on silica gel (60−20 mesh) using petroleum ether−EtOAc (9:1) as eluent to furnish 8 (3.6 g, 91%) as a colorless oil.  $[\alpha]_{\text{D}}^{25} - 61$  (c 0.27, CHCl<sub>3</sub>); <sup>1</sup>H NMR (CDCl<sub>3,</sub> 300 MHz): δ 1.31 (s, 3H), 1.47 (s, 3H), 2.33 (s, 3H), 3.12 (dd, 1H, J  $= 6.9, 13.5 \text{ Hz}$ ), 3.26 (dd, 1H, J = 7.2, 13.5 Hz), 3.81 (s, 3H), 3.88 (d, 1H,  $J = 3.0$  Hz), 4.25 (dt, 1H,  $J = 3.0$ , 6.9 Hz), 4.44 (d, 1H,  $J = 11.4$ Hz), 4.49 (merged d, 1H), 4.61 (d, 1H,  $J = 11.1$  Hz), 5.90 (d, 1H,  $J =$ 3.6 Hz), 6.89 (d, 2H,  $J = 8.4$  Hz), 7.26 (d, 2H,  $J = 8.4$  Hz); <sup>13</sup>C NMR (CDCl<sub>3</sub>, 75 MHz):  $\delta$  26.2 (CH<sub>3</sub>), 26.7 (CH<sub>3</sub>), 27.0 (CH<sub>2</sub>), 30.4  $(CH<sub>3</sub>)$ , 55.2 (CH<sub>3</sub>), 71.7 (CH<sub>2</sub>), 79.1 (CH), 81.5 (CH), 82.1 (CH), 105.0 (CH), 111.6 (C), 113.8 (2 × CH), 129.2 (C), 129.4 (2 × CH), 159.4 (C), 195.2 (C); HRMS (ESI−QToF, positive ion) calcd for  $C_{18}H_{24}NaO_6S$ ,  $m/z$  391.1191, found 391.1187.

(3aR,5S,6R,6aR)-5-(Allylthiomethyl)-6-(4-methoxybenzyloxy)-2,2-dimethyltetrahydrofuro[2,3-d][1,3]dioxole (9). To a solution of 8 (2.0 g, 5.43 mmol) in MeOH (50 mL) at 0  $^{\circ}$ C was added NaBH4 (619 mg, 16.29 mmol) portionwise. After 10 min of stirring, allyl bromide (0.95 mL, 10.86 mmol) was added to the mixture by a syringe. A methanolic solution of NaOMe (28 wt %) (2.7 mL, 10.8 mmol) was added dropwise to the mixture, which was allowed to stir at room temperature for 3 h. The solvent was evaporated in vacuo, and to the residue EtOAc (25 mL) and water (30 mL) were added. The organic layer was separated, and the aqueous layer was extracted with EtOAc  $(2 \times 10 \text{ mL})$ . The combined organic layer was washed with H<sub>2</sub>O ( $2 \times 10$  mL) and dried (Na<sub>2</sub>SO<sub>4</sub>), and the solvent was removed in vacuo. The crude product was purified by column chromatography over silica gel (100−200 mesh) using petroleum ether−EtOAc (9:1) as eluent to obtain 9 (1.44 g, 72%) as a yellow oil.  $[\alpha]_{\rm D}^{25}$  – 68 (c 0.24, CHCl<sub>3</sub>); <sup>1</sup>H NMR (CDCl<sub>3,</sub> 300 MHz):  $\delta$  1.32 (s, 3H), 1.49 (s, 3H), 2.77 (d, 2H, J = 7.2 Hz), 3.17 (d, 2H, J = 6.9 Hz), 3.81 (s, 3H), 3.96 (d, 1H, J = 3.0 Hz), 4.31 (dt, 1H, J  $= 3.0, 7.2$  Hz), 4.47 (d, 1H,  $J = 3.9$  Hz), 4.58 (d, 1H,  $J = 3.9$  Hz), 4.61  $(d, 1H, J = 11.7 Hz)$ , 5.07–5.13 (m, 2H), 5.75–5.86 (m, 1H), 5.90 (d, 1H, J = 3.6 Hz), 6.88 (d, 2H, J = 8.7 Hz), 7.26 (d, 2H, J = 8.4 Hz); <sup>13</sup>C NMR (CDCl<sub>3,</sub> 75 MHz):  $\delta$  26.2 (CH<sub>3</sub>), 26.7 (CH<sub>3</sub>), 28.2 (CH<sub>2</sub>), 35.4  $(CH<sub>2</sub>)$ , 55.2 (CH<sub>3</sub>), 71.8 (CH<sub>2</sub>), 79.9 (CH), 81.3 (CH), 82.0 (CH), 105.0 (CH), 111.4 (C), 113.7 (2  $\times$  CH), 117.2 (CH<sub>2</sub>), 129.35 (2  $\times$ CH), 129.41 (C), 134.2 (CH), 159.3 (C); HRMS (ESI−QToF, positive ion) calcd for  $C_{19}H_{26}NaO_5S$ ,  $m/z$  389.1399, found 389.1414.

(3aR,5S,6R)-5-(Allylthiomethyl)-2,2-dimethyltetrahydrofuro- [2,3-d][1,3]dioxol-6-ol (10). Deprotection of the PMB group was carried out, following the procedure as described for the preparation of 4, using 9 (1.2 g, 3.28 mmol) and DDQ (1.12 g, 4.92 mmol). The usual work up and purification by column chromatography on silica gel (100−200 mesh) using petroleum ether−EtOAc (17:3) furnished 10 (0.65 g, 80%) as a colorless oil.  $[\alpha]_{D}^{25} - 40$  (c 0.27, CHCl<sub>3</sub>); <sup>1</sup>H NMR  $(CDCI<sub>3</sub> 300 MHz): \delta 1.32$  (s, 3H), 1.50 (s, 3H), 2.15 (d, 1H, J = 5.4 Hz), 2.74 (dd, 1H,  $J = 9.0$ , 13.2 Hz), 2.86 (dd, 1H,  $J = 5.4$ , 13.2 Hz), 3.21 (d, 2H, J = 7.2 Hz), 4.26–4.33 (m, 2H), 4.53 (d, 1 H, J = 3.6 Hz), 5.12−5.88 (m, 2H), 5.74−5.88 (m, 1H), 5.92 (d, 1H,  $J = 3.3$  Hz); <sup>13</sup>C NMR (CDCl<sub>3</sub>, 75 MHz):  $\delta$  26.1 (CH<sub>3</sub>), 26.6 (CH<sub>3</sub>), 27.9 (CH<sub>2</sub>), 35.3 (CH<sub>2</sub>), 74.8 (CH), 79.4 (CH), 84.9 (CH), 104.6 (CH), 111.6 (C), 117.6 (CH2), 133.9 (CH); HRMS (ESI−QToF, positive ion) calcd for  $C_{11}H_{18}NaO_4S$ ,  $m/z$  269.0823, found 269.0836.

(3S,6aS,7aR,10aR,10bS)-1-Benzyl-9,9-dimethylhexahydro-1H-3,10b-methano[1,3]dioxolo[4′,5′:4,5]furo[3,2-c][1,6,2]oxathiazocine (11). To a solution of oxalyl chloride (0.35 mL, 4.06 mmol) in dry DCM (7 mL) cooled to −65 °C was added a solution of dry DMSO (0.52 mL, 7.27 mmol) in DCM (2 mL) dropwise under  $N_2$ , and the mixture was stirred for 15 min. A solution of 10 (0.500 g, 2.03 mmol) in DCM (5 mL) was added to the above mixture over a

period of 1 h, and the stirring was continued for another 1 h.  $Et<sub>3</sub>N$  (3 mL) was added to it, and the reaction mixture was allowed to reach room temperature. After quenching the reaction with addition of water (5 mL), the mixture was extracted with DCM (3  $\times$  30 mL). The combined extract was washed with water  $(2 \times 30 \text{ mL})$  and dried  $(Na_3SO_4)$ , and the solvent was evaporated in vacuo to furnish a crude ketone, which was treated with BnNHOH (0.375 g, 3.05 mmol) in refluxing toluene (20 mL) for 10 h. The usual work up and purification by column chromatography on silica gel (230−400 mesh) and elution with petroleum ether−EtOAc (17:3) furnished 11 (0.300 g, 42%) as a yellow viscous liquid.  $[\alpha]_{\text{D}}^{25}$  + 129 (c 0.21, CHCl<sub>3</sub>); <sup>1</sup>H NMR (CDCl<sub>3,</sub> 600 MHz):  $\delta$  1.35 (s, 3H), 1.60 (s, 3H), 2.22 (apparent t, 1H, J = 9.6, 11.4 Hz), 2.40 (dd, 1H,  $J = 4.2$ , 14.4 Hz), 2.86 (brd, 1H,  $J = 15.6$  Hz), 2.99 (brd, 1H, J = 15.0 Hz), 3.22 (dd, 1H, J = 2.4,16.8 Hz), 3.28–3.29 (brs, 1H), 3.88−3.90 (brs, 1H), 4.20 (brd, 1H, J = 12.0 Hz), 4.47− 4.52 (m, 1H), 4.55 (d, 1H,  $J = 3.6$  Hz), 5.02 (brs, 1H), 5.93 (d, 1H,  $J =$ 3.0 Hz), 7.28-7.40 (m, 5H); <sup>13</sup>C NMR (CDCl<sub>3,</sub> 75 MHz): δ 26.2  $(CH_3)$ , 26.5 (CH<sub>3</sub>), 30.4 (CH<sub>2</sub>), 33.1 (CH<sub>2</sub>), 34.7 (CH<sub>2</sub>), 56.7 (CH<sub>2</sub>), 77.2 (CH), 77.9 (CH), 82.6 (CH), 103.8 (CH), 113.1 (C), 127.3 (CH), 128.3 (2  $\times$  CH), 129.1 (2  $\times$  CH), 137.7 (C), one (C) not discernible; HRMS (ESI−QToF, positive ion) calcd for  $C_{18}H_{23}NNaO_4S$ ,  $m/z$  372.1245, found 372.1241.

((3aR,5R,6S,6aR)-6-(4-Methoxybenzyloxy)-2,2-dimethyltetrahydrofuro[2,3-d][1,3]dioxol-5-yl)methylmethanesulfonate (12).  $CH_3SO_2Cl$  (1.3 mL, 17.12 mmol) was added to an ice−cold solution of 2 (3.32 g, 10.7 mmol) in DCM (50 mL). After stirring for 5 min,  $Et<sub>3</sub>N$  (2.25 mL, 16 mmol) was added dropwise to the mixture and it was stirred at room temperature for 2 h. The organic layer was washed with a saturated solution of NaHCO<sub>3</sub> (3  $\times$ 10 mL), and water  $(3 \times 10 \text{ mL})$ , and then dried  $(Na_2SO_4)$ , and evaporated to a residue, which was purified by column chromatography on silica gel (60−120 mesh) using petroleum ether−EtOAc (9:1) as eluent to furnish 12 (3.92 g, 94%) as a colorless liquid.  $[\alpha]_{\text{D}}^{25}$  $-$  28 (c 0.38, CHCl<sub>3</sub>); <sup>1</sup>H NMR (CDCl<sub>3,</sub> 300 MHz):  $\delta$  1.33 (s, 3H), 1.49 (s, 3H), 3.03 (s, 3H), 3.81 (s, 3H), 3.98 (brs, 1H), 4.34−4.44 (m, 4H), 4.60−4.63 (m, 2H), 5.95 (d, 1H, J = 3.6 Hz), 6.89 (d, 2H, J = 8.4 Hz), 7.23 (d, 2H, J = 8.4 Hz); <sup>13</sup>C NMR (CDCl<sub>3</sub>, 75 MHz):  $\delta$  26.2 (CH<sub>3</sub>), 26.8 (CH<sub>3</sub>), 37.05 (CH<sub>3</sub>), 55.3 (CH<sub>3</sub>), 67.8 (CH<sub>2</sub>), 71.6 (CH<sub>2</sub>), 77.8 (CH), 80.9 (CH), 81.9 (CH), 105.3 (CH), 112.0 (C), 114.0 (2 × CH), 128.8 (C), 129.6 (2 × CH), 159.6 (C); HRMS (EI, magnetic sector, positive ion) calcd for  $C_{17}H_{24}O_8S$ ,  $m/z$  388.1192, found 388.1196.

(3aR,5R,6S,6aR)-5-(Azidomethyl)-6-(4-methoxybenzyloxy)- 2,2-dimethyltetrahydrofuro[2,3-d][1,3]dioxole (13). A mixture of 12 (4.3 g, 11.1 mmol) and  $\text{NaN}_3$  (6.63 g, 102.0 mmol) in anhydrous DMF (50 mL) was heated at 80−100 °C for 6 h. The reaction mixture was cooled to room temperature, and the solvent was evaporated under reduced pressure. The residue was extracted with CHCl<sub>3</sub> ( $3 \times 30$  mL); the combined extract was washed with water, dried  $(Na<sub>2</sub>SO<sub>4</sub>)$ , and evaporated in vacuo. The crude product was chromatographically purified on silica gel (100−200 mesh) using petroleum ether−EtOAc (19:1) as eluent to give 13 (3.51 g, 94%) as a colorless liquid.  $[\alpha]_{\text{D}}^{25} - 37$  (c 0.24, CHCl<sub>3</sub>), <sup>1</sup>H NMR (CDCl<sub>3,</sub> 300 MHz):  $\delta$  1.33 (s, 3H), 1.50 (s, 3H), 3.45 (dd, 1H, J = 6.6, 12.3 Hz), 3.56 (dd, 1H,  $J = 6.9$ , 12.3 Hz), 3.81 (s, 3H), 3.92 (d, 1H,  $J = 3.3$  Hz), 4.29 (td, 1H,  $J = 3.3$ , 6.3 Hz), 4.44 (d, 1H,  $J = 11.4$  Hz), 4.61 (d, 1H,  $J$  $= 3.9$  Hz), 4.62 (d, 1H, J = 11.4 Hz), 5.92 (d, 1H, J = 3.6 Hz), 6.89 (d, 2H,  $J = 8.4$  Hz), 7.24 (d, 2H, merged with CDCl<sub>3</sub>); <sup>13</sup>C NMR (CDCl<sub>3</sub> 75 MHz):  $\delta$  26.1 (CH<sub>3</sub>), 26.7 (CH<sub>3</sub>), 49.1 (CH<sub>2</sub>), 55.1 (CH<sub>3</sub>), 71.4  $(CH<sub>2</sub>)$ , 78.6 (CH), 80.9 (CH), 81.9 (CH), 104.9 (CH), 111.7 (C), 113.8 (2 × CH), 128.9 (C), 129.4 (2 × CH), 159.4 (C); HRMS (ESI–QToF, positive ion) calcd for C<sub>16</sub>H<sub>21</sub>N<sub>3</sub>NaO<sub>5</sub>, m/z 358.1379, found 358.1370.

((3aR,5R,6S,6aR)-6-(4-Methoxybenzyloxy)-2,2-dimethyltetrahydrofuro[2,3-d][1,3]dioxol-5-yl)methanamine (14). To a mixture of 13 (2.35 g, 7.01 mmol) and Ph<sub>3</sub>P (2.75 g, 10.52 mmol) taken in THF (30 mL) was added water (0.20 mL, 10.5 mmol), and the reaction mixture was heated at 110 °C for 6 h. The mixture was extracted with Et<sub>2</sub>O ( $3 \times 50$  mL), and the combined organic extract was washed with water (100 mL) and brine (100 mL). The solvent was dried  $(Na_2SO_4)$  and concentrated to a crude yellow oil, which was purified by column chromatography on silica gel (100−200 mesh) using petroleum ether−EtOAc (1:1) as eluent to furnish 14 (1.54 g, 71%) as a colorless liquid.  $[\alpha]_{\rm D}^{25} - 38$  (c 0.28, CHCl<sub>3</sub>); <sup>1</sup>H NMR  $(CDCl<sub>3</sub> + D<sub>2</sub>O<sub>3</sub> 300 MHz): \delta 1.33 (s, 3H), 1.49 (s, 3H), 2.88 (dd, 1H, 1.49)$  $J = 5.1, 13.2 \text{ Hz}$ ), 3.01 (dd, 1H,  $J = 6.3, 13.2 \text{ Hz}$ ), 3.81 (s, 3H), 3.89 (d, 1H,  $J = 3.3$  Hz), 4.13 (brd, 1H,  $J = 3.0$  Hz), 4.39 (d, 1H,  $J = 11.7$  Hz), 4.62 (d, 1H,  $J = 3.6$  Hz), 4.65 (d, 1H,  $J = 11.7$  Hz), 5.94 (d, 1H,  $J = 3.6$ Hz), 6.89 (d, 2H,  $J = 8.7$  Hz), 7.25 (merged d, 2H); <sup>13</sup>C NMR (CDCl<sub>3,</sub> 75 MHz):  $\delta$  26.2 (CH<sub>3</sub>), 26.7 (CH<sub>3</sub>), 40.2 (CH<sub>2</sub>), 55.2  $(CH<sub>3</sub>), 71.3 (CH<sub>2</sub>), 80.8 (CH), 81.3 (CH), 82.3 (CH), 104.9 (CH),$ 111.5 (C), 113.9 (2  $\times$  CH), 129.2 (C), 129.5 (2  $\times$  CH), 159.4 (C); HRMS (ESI–QToF, positive ion) calcd for C<sub>16</sub>H<sub>23</sub>NNaO<sub>5</sub>,  $m/z$ 332.1474, found 332.1473.

O-tert-Butyl-(((3aR,5R,6S,6aR)-6-(4-Methoxybenzyloxy)-2,2 dimethyltetrahydrofuro[2,3-d][1,3]dioxol-5-yl)methyl)carbamate (15). To a solution of 14 (1.50 g, 4.85 mmol) in DCM (20 mL) was added di-tert-butyl dicarbonate (1.10 mL, 4.85 mmol), and the mixture was stirred at room temperature for 4 h. The mixture was diluted with CHCl<sub>3</sub> (20 mL) and then extracted with CHCl<sub>3</sub> ( $3 \times 50$ ) mL). The combined organic extract was washed with water (50 mL) and brine (50 mL), and dried  $(Na_2SO_4)$  and concentrated under reduced pressure to give a crude residue, which was purified by column chromatography on silica gel (60−120 mesh) using petroleum ether− EtOAc  $(4:1)$  as eluent to furnish 15  $(1.6 g, 81\%)$  as a colorless liquid.  $[\alpha]_D^{25}$  – 33 (c 0.47, CHCl<sub>3</sub>); <sup>1</sup>H NMR (CDCl<sub>3,</sub> 300 MHz):  $\delta$  1.32 (s, 3H), 1.42 (s, 9H), 1.48 (s, 3H), 3.34 (brs, 1H), 3.49 (brs, 1H), 3.81 (s, 3H), 3.88 (s, 1H), 4.22 (brs, 1H), 4.39 (d, 1H, J = 11.4 Hz), 4.60 (s, 1H), 4.62 (d, 1H, merged), 4.73 (brs, 1H), 5.92 (s, 1H), 6.89 (d, 2H, J  $= 7.5$  Hz), 7.25 (d, 2H, merged). <sup>13</sup>C NMR (CDCl<sub>3</sub>, 75 MHz):  $\delta$  26.2 (CH<sub>3</sub>), 26.7 (CH<sub>3</sub>), 28.3 (3  $\times$  CH<sub>3</sub>), 39.5 (CH<sub>2</sub>), 55.2 (CH<sub>3</sub>), 71.4  $(CH<sub>2</sub>)$ , 79.1 (CH), 81.2 (CH), 82.3 (CH), 104.9 (CH), 111.6 (C), 114.0 (2 × CH), 129.1 (C), 129.4 (2 × CH), 155.9 (C), 159.4 (C), one quaternary C was not discernible; HRMS (ESI−QToF, positive ion) calcd for  $C_{21}H_{31}NNaO_7$ ,  $m/z$  432.1998, found 432.2017.

O-tert-Butylallyl(((3aR,5R,6S,6aR)-6-((4-methoxybenzyl)oxy)- 2,2-dimethyltetrahydrofuro[2,3-d][1,3]dioxol-5-yl)methyl) carbamate (16). To a solution of 15 (3.10 g, 7.58 mmol) in dry DMF (80 mL) was added TBAB (0.245 g, 0.76 mmol), and the mixture was stirred at 0  $^{\circ}$ C with portionwise addition of NaH (60% in mineral oil, 0.394 g, 9.85 mmol). After 10 min, allyl bromide (0.85 mL, 9.85 mmol) was added to it, and the resulting mixture was stirred overnight at room temperature. The solvent was evaporated under reduced pressure. The residue was extracted with CHCl<sub>3</sub> ( $3 \times 30$  mL); the combined extract was washed with water, dried  $(Na<sub>2</sub>SO<sub>4</sub>)$ , and evaporated in vacuo. The crude product was purified by chromatography on silica gel (100−200 mesh) using petroleum ether−EtOAc  $(9.1)$  as eluent to furnish 16 (3.0 g, 88%) as colorless liquid.  $[\alpha]_D^{25} + 3$  $(c$  0.15, CHCl<sub>3</sub>); <sup>1</sup>H NMR (CDCl<sub>3,</sub> 300 MHz):  $\delta$  1.31 (s, 3H), 1.43 (s, 9H), 1.47 (s, 3H), 3.19−3.44 (m, 2H), 3.81 (s, 3H), 3.91−4.05 (m, 3H), 4.38 (brs, 1H), 4.40 (d, 1H, J = 11.7 Hz), 4.57 (brs, 2H), 5.03− 5.09 (m, 2H), 5.70−5.80 (m, 1H), 5.92 (s, 1H), 6.88 (d, 2H, J = 7.8 Hz), 7.23 (d, 2H, J = 8.4 Hz); <sup>13</sup>C NMR (CDCl<sub>3</sub>, 75 MHz):  $\delta$  26.2 (CH<sub>3</sub>), 26.7 (CH<sub>3</sub>), 28.3 (3  $\times$  CH<sub>3</sub>), 46.4 (CH<sub>2</sub>), 50.7 (CH<sub>2</sub>), 55.2  $(CH_3)$ , 71.5  $(CH_2)$ , 79.5  $(C)$ , 79.9  $(CH)$ , 81.9  $(CH)$ , 82.0  $(CH)$ , 105.0 (CH), 111.4 (C), 113.9 (2  $\times$  CH), 115.6 (CH<sub>2</sub>), 129.4 (2  $\times$ CH), 134.1 (CH), 155.8 (C), 159.4 (C), one quaternary C not discernible; HRMS (ESI−QToF, positive ion) calcd for  $C_{24}H_{35}NNaO_7$ ,  $m/z$  472.2311, found 472.2299.

O-tert-Butylallyl-(((3aR,5R,6S,6aR)-6-hydroxy-2,2-dimethyltetrahydrofuro[2,3-d][1,3]dioxol-5-yl)methyl)carbamate (17). Removal of PMB protection was done, according to the procedure as adopted in the preparation of 4, using 16 (2.0 g, 4.45 mmol) and DDQ (1.52 g, 6.68 mmol). The usual work up and purification by column chromatography on silica gel (100−200 mesh) using petroleum ether−EtOAc (5:1) afforded 17 (1.2 g, 82%) as a colorless oil:  $[\alpha]_D^{25}$  + 49 (c 0.23, CHCl<sub>3</sub>); <sup>1</sup>H NMR (CDCl<sub>3</sub> + D<sub>2</sub>O, 300 MHz):  $\delta$  1.32 (s, 3H), 1.45 (s, 9H), 1.51 (s, 3H), 3.05 (brd, 1H, J = 13.8 Hz), 3.62 (brdd, 1H, J = 5.1, 15.6 Hz), 3.86 (t, 1H, J = 12.3 Hz), 3.95−4.07 (m, 2H), 4.12 (brd, 1H, J = 10.2 Hz), 4.60 (s, 1H), 5.12−5.18 (m,

2H), 5.72–5.77 (m, 1H), 5.92 (s, 1H); <sup>13</sup>C NMR (CDCl<sub>3</sub>, 75 MHz):  $\delta$  26.0 (CH<sub>3</sub>), 26.7 (CH<sub>3</sub>), 28.2 (3  $\times$  CH<sub>3</sub>), 44.0 (CH<sub>2</sub>), 51.3 (CH<sub>2</sub>), 73.7 (CH), 78.5 (CH), 81.2 (C), 84.7 (CH), 104.7 (CH), 111.2 (C), 117.1 (CH2), 133.0 (CH), 156.9 (C); HRMS (ESI−QToF, positive ion) calcd for  $C_{16}H_{27}NNaO_6$ ,  $m/z$  352.1736, found 352.1720.

(3S,6aR,7aR,10aR,10bR)-tert-Butyl-1-benzyl-9,9-dimethylhexahydro-3,10b-methano[1,3]dioxolo[4′,5′:4,5]furo[3,2-c]- [1,2,6]oxadiazocine-5(1H)-carboxylate (18). Oxidation of 17  $(2.2)$ g, 6.69 mmol) in dry DCM (30 mL) was carried out using DMP (4.26 g, 10.04 mmol) as described in the preparation of 6. The usual work up afforded a crude ketone, which was treated with BnNHOH (1.23 g, 10.04 mmol) in refluxing toluene (50 mL) for 12 h. The usual work up and purification by column chromatography on silica gel (230−400 mesh) using petroleum ether−EtOAc (17:3) furnished 18 (1.85 g, 64%) as a colorless solid. mp 167–168 °C;  $[\alpha]_D^{25}$  + 102 (c 0.18, CHCl<sub>3</sub>); <sup>1</sup>H NMR (CDCl<sub>3,</sub> 300 MHz):  $\delta$  1.34 (s, 3H), 1.46 (s, 9H), 1.60 (s, 3H), 2.19−2.36 (m, 2H), 2.91 (brs,1H), 3.21 (brs,1H), 3.88  $(m, 1H)$ , 4.14−4.23  $(m, 4H)$ , 4.51  $(d, 1H, J = 3.0 Hz)$ , 4.76  $(brs, 1H)$ , 5.83 (d, 1H, J = 3.3 Hz), 7.29–7.41 (m, 5H); <sup>13</sup>C NMR (CDCl<sub>3,</sub> 75 MHz):  $\delta$  26.1 (CH<sub>3</sub>), 26.5 (CH<sub>3</sub>), 28.2 (3 × CH<sub>3</sub>), 32.9 (CH<sub>2</sub>), 45.4 (CH<sub>2</sub>), 49.4 (CH<sub>2</sub>), 57.5 (CH<sub>2</sub>), 76.2 (CH), 77.2 (CH), 78.6 (C), 80.0 (C), 82.3 (CH), 103.8 (CH), 112.7 (C), 127.3 (CH), 128.3 (2 × CH), 129.2 (2 × CH), 137.8 (C), 156.2 (C); HRMS (EI, magnetic sector, positive ion) calcd for  $C_{23}H_{32}N_2O_6$ ,  $m/z$  432.2260, found 432.2250.

2-((3aR,5R,6S,6aR)-6-(4-Methoxybenzyloxy)-2,2-dimethyltetrahydrofuro[2,3-d][1,3]dioxol-5-yl)ethanol (20). A solution of 19 (4.75g, 15.52 mmol) in THF (40 mL) at 0 °C was treated with a solution of 9-BBN (94 mL, 0.5 M in THF, 46.57 mmol), the reaction mixture was allowed to warm up to 20 °C and stirred overnight. The mixture was cooled in an ice−water bath, treated carefully with an aqueous solution of sodium hydroxide (2 N, 50 mL), followed by hydrogen peroxide (50 mL, 30% in water), warmed up to 20 °C, and stirred for 2 h more. The solvent was removed in vacuum to afford a residue, which was partitioned between ether (100 mL) and water (100 mL). The separated ether layer was dried  $(Na_2SO_4)$  and concentrated. Chromatographic purification of the residue on silica gel (100−200 mesh) using petroleum ether−EtOAc (3:1) as eluent provided 20 (4.0 g, 80%) as a colorless oil.  $[\alpha]_{\rm D}^{25} - 28$  (c 0.12, CHCl<sub>3</sub>); <sup>1</sup>H NMR (CDCl<sub>3</sub>, 300 MHz):  $\delta$  1.32 (s, 3H), 1.48 (s, 3H), 1.73−1.87 (m, 1H), 1.98−2.10 (m, 1H), 2.25 (brs, 1H), 3.73 (t, 2H, J  $= 5.7$  Hz), 3.80 (s, 3H), 4.28–4.34 (m, 1H), 4.41 (d, 1H, J = 11.7 Hz), 4.61 (d, 1H,  $J = 4.2$  Hz), 4.63 (d, 1H,  $J = 12.9$  Hz), 5.91 (d, 1H,  $J = 3.9$ Hz), 6.87 (d, 2H, J = 8.7 Hz), 7.24 (d, 2H, J = 8.4 Hz), one H not discernible; <sup>13</sup>C NMR (CDCl<sub>3</sub>, 75 MHz):  $\delta$  26.1 (CH<sub>3</sub>), 26.5 (CH<sub>3</sub>), 30.8 (CH<sub>2</sub>), 55.2 (CH<sub>3</sub>), 60.1 (CH<sub>2</sub>), 71.3 (CH<sub>2</sub>), 78.4 (CH), 81.9 (CH), 82.1 (CH), 104.6 (CH), 111.3 (C), 113.8 (2 × CH), 129.3 (2 × CH), 159.3 (C), one C not discernible; HRMS (ESI−QToF, positive ion) calcd for  $C_{17}H_{24}NaO_6$ ,  $m/z$  347.1471, found 347.1463.

1-((3aR,5R,6S,6aR)-6-(4-Methoxybenzyloxy)-2,2-dimethyltetrahydrofuro[2,3-d][1,3]dioxol-5-yl)pent-4-en-2-ol (21). To a solution of 20 (2.3 g, 7.10 mmol) in DCM (40 mL) at  $0^{\circ}$ C was added DMP (4.34 g, 10.23 mmol) under  $N_2$ , and the reaction was stirred for 2 h. The residue was extracted with DCM  $(2 \times 40 \text{ mL})$ , and the solvent was washed with a saturated solution of NaHCO<sub>3</sub> (50 mL), 10%  $Na<sub>2</sub>S<sub>2</sub>O<sub>3</sub>$  solution (50 mL), and brine (50 mL). The organic solvent was dried  $(Na_2SO_4)$  and evaporated in vacuo to give an aldehyde. To the crude aldehyde dissolved in NH4Cl−THF (5:1, 36 mL) at 0 °C was added allyl bromide (1.84 mL, 21.3 mmol), and the reaction mixture was stirred for 5 min. Zn dust (2.7 g,41.2 mmol) was added portionwise to the reaction mixture, which was stirred at room temperature for 12 h. The reaction was quenched with a saturated aqueous NaHCO<sub>3</sub> solution (50 mL), and the solvent was evaporated to furnish a residue, which was extracted with CHCl<sub>3</sub> ( $3 \times 40$  mL). The combined extract was washed with brine (50 mL), dried  $(Na<sub>2</sub>SO<sub>4</sub>)$ , and evaporated in vacuo. The crude product was purified by column chromatography on silica gel (100−200 mesh) using petroleum ether−EtOAc (8:2) to furnish 21 (2.0 g, 77%) as a colorless liquid (a mixture of  $\alpha$  and  $\beta$  anomers). [ $\alpha$ ] $_{\rm D}^{25}$  – 36 (c 0.35, CHCl<sub>3</sub>); <sup>1</sup>H NMR (CDCl<sub>3</sub>, 300 MHz): δ 1.32 (s), 1.48 (s), 1.61–1.78 (m), 1.81–

2.05 (m), 2.15−2.35 (m), 3.72−3.77 (m), 3.81 (s), 3.84−3.92 (m), 4.33−4.37 (m), 4.41 (dd, J = 6.3, 11.7 Hz), 4.60 (q, J = 3.6 Hz), 4.65  $(d, J = 3.6 \text{ Hz})$ , 5.14 (m), 5.74–5.88 (m), 5.9 (d, J = 4.2 Hz), 5.92 (d, J  $= 4.2$  Hz), 6.88 (d, J = 8.7), 7.25 (d, J = 7.8 Hz); <sup>13</sup>C NMR (CDCl<sub>3</sub>, 75 MHz):  $\delta$  26.0 (CH<sub>3</sub>), 26.4 (CH<sub>3</sub>), 34.4 (CH<sub>2</sub>), 34.8 (CH<sub>2</sub>), 41.8  $(CH<sub>2</sub>)$ , 42.3 (CH<sub>2</sub>), 55.0 (CH<sub>3</sub>), 67.7 (CH), 69.1 (CH<sub>3</sub>), 71.1 (CH<sub>2</sub>), 77.2 (CH), 78.8 (CH), 81.5 (CH), 81.8 (CH), 82.2 (CH), 104.3 (CH), 104.5 (CH), 111.0 (C), 111.2 (C), 113.6 (CH), 117.3 (CH<sub>2</sub>), 117.5 (CH<sub>2</sub>), 129.2 (CH), 129.4 (CH), 134.6 (CH), 159.1 (C); HRMS (ESI-QToF, positive ion) calcd for  $C_{20}H_{28}NaO_6$ ,  $m/z$ 387.1784, found 387.1781.

1-((3aR,5R,6S,6aR)-6-Hydroxy-2,2-dimethyltetrahydrofuro- [2,3-d][1,3]dioxol-5-yl)pent-4-en-2-yl acetate (22). To a solution of 21 (1.38 g, 3.8 mmol) in pyridine (20 mL) were added  $Ac_2O$  (0.72 mL, 7.58 mmol) and DMAP (pinch), and the mixture was stirred at room temperature for 8 h. Pyridine was evaporated by azeotropic distillation with toluene in rotary evaporator. The residue was extracted with DCM  $(2 \times 50 \text{ mL})$ , and the solvent was washed with brine (50 mL), dried ( $Na<sub>2</sub>SO<sub>4</sub>$ ), and evaporated. The residue was purified by column chromatography on silica gel (60−120 mesh) using petroleum ether−EtOAc (1:4) as eluent to furnish the triacetylated derivative (1.48 g, 96%) as a thick oil. The removal of the PMB protection was done, following the method as described in 4, using the oil (1.08 g, 2.66 mmol) and DDQ (0.91 g, 4.01 mmol) in a mixture of DCM (40 mL) and  $H_2O$  (2 mL). The usual work up and purification by column chromatography on silica gel (100−200 mesh) using petroleum ether−EtOAc (5:1) as eluent furnished 22 (685 mg, 90%) as a colorless oil.  $[\alpha]_{D}^{25} - 10$  (c 0.55, CHCl<sub>3</sub>); <sup>1</sup>H NMR (CDCl<sub>3</sub>, 300 MHz): δ 1.31 (s), 1.48 (s), 1.88−2.03 (m), 2.05 and 2.06 (2s), 2.37− 2.45 (m),  $4.07 - 4.22$  (m),  $4.49$  and  $4.54$  ( $2 \times d$ ,  $J = 3.6$  Hz),  $4.83 - 4.92$ (quint,  $J = 6.0$  Hz),  $5.04 - 5.14$  (m),  $5.69 - 5.83$  (m),  $5.88$  (t,  $J = 3.9$ Hz); <sup>13</sup>C NMR (CDCl<sub>3</sub>, 75 MHz):  $\delta$  21.0 (CH<sub>3</sub>), 26.0 (CH<sub>3</sub>), 26.5  $(CH<sub>3</sub>), 31.5 (CH<sub>2</sub>), 31.9 (CH<sub>2</sub>), 38.6 (CH<sub>2</sub>), 38.8 (CH<sub>2</sub>), 70.8 (CH<sub>2</sub>),$ 71.3 (CH), 75.0 (CH), 75.3 (CH), 77.0 (CH), 77.3 (CH), 85.0 (CH), 85.2 (CH), 104.0 (CH), 104.1 (CH), 111.18 (C), 111.21 (C), 117.9  $(CH<sub>2</sub>)$ , 118.1 (CH<sub>2</sub>), 132.9 (CH), 133.1 (CH), 170.9 (C); HRMS (ESI–QToF, positive ion) calcd for C<sub>14</sub>H<sub>22</sub>NaO<sub>6</sub>, m/z 309.1314, found 309.1307.

(3S,5R,6aR,7aR,10aR,10bR)-1-Benzyl-9,9-dimethyloctahydro-3,10b-methano[1,3]dioxolo[4′,5′:4,5]furo[3,2-c][1,2] oxazocin-5-yl acetate (23). Oxidation of the hydroxyl group of 22, followed by nitrone cycloaddition reaction, was carried out, according to the procedure as described in 6, using 22 (0.64 g, 2.24 mmol) and DMP (1.42 g, 3.4 mmol) for oxidation, and BnNHOH (0.42 g, 3.41 mmol) in refluxing toluene (10 mL) for 10 h for cycloaddition reaction. The usual work up and purification by column chromatography on silica gel (230−400 mesh) using petroleum ether−EtOAc (10:1.5) furnished 23 (0.35 g, 40%) as a colorless solid. mp 186−187  $^{\circ}$ C; [ $\alpha$ ]<sup>25</sup> – 80 (c 0.11, CHCl<sub>3</sub>); <sup>1</sup>H NMR (CDCl<sub>3</sub>, 600 MHz):  $\delta$  1.34 (s, 3H), 1.60 (s, 3H), 1.58−1.62 (a merged signal 1H), 2.00 (s, 3H), 2.13−2.31 (m, 5H), 3.81 (brs, 1H), 4.16 (d, 1H, J = 13.2 Hz), 4.43  $(brs, 1H)$ , 4.49 (d, 1H,  $J = 3.6$  Hz), 4.69 (brs, 1H), 5.19 (dt, 1H,  $J =$ 5.4,10.2 Hz), 5.83 (d, 1H,  $J = 3.6$  Hz), 7.27 (t, 1H,  $J = 7.2$  Hz), 7.34 (t, 2H, J = 7.8 Hz), 7.40 (d, 2H, J = 8.4 Hz); <sup>13</sup>C NMR (CDCl<sub>3</sub>, 150 MHz):  $\delta$  21.3 (CH<sub>3</sub>), 26.2 (CH<sub>3</sub>), 26.4 (CH<sub>3</sub>), 33.5 (CH<sub>2</sub>), 37.4 (CH<sub>2</sub>), 57.4 (CH<sub>2</sub>), 66.8 (CH), 73.8 (CH), 82.2 (CH), 103.7 (CH), 113.0 (C), 127.3 (CH), 128.3 (2 × CH), 129.2 (2 × CH), 137.7 (C),  $170.2$  (C), one CH<sub>2</sub>, one CH and one C not discernible; HRMS (EI, magnetic sector, positive ion) calcd for  $C_{21}H_{27}NO_6$ ,  $m/z$  389.1838, found 389.1841.

(3aR,5S,6S,6aR)-5-((R)-1-(Benzyloxy)but-3-en-1-yl)-2,2 dimethyltetrahydrofuro[2,3-d][1,3]dioxol-6-ol (25). Sodium metaperiodate (3.89 g, 18.2 mmol) dissolved in water (10 mL) was added to a solution of  $24$  (5.32 g, 15.2 mmol) in MeOH (150 mL) at 0 °C, and the reaction was stirred for 3 h. The reaction mixture was filtered, and the residue was washed with MeOH  $(2 \times 30 \text{ mL})$ . The combined filtrate was evaporated to furnish a residue, which was extracted with DCM  $(2 \times 50 \text{ mL})$  and washed with brine  $(30 \text{ mL})$ . The solvent was dried  $(Na_2SO_4)$  and evaporated in vacuo to give a crude aldehyde (4.22 g). The aldehyde (3.5 g) was allylated using allyl bromide (2.95 mL, 34.1 mmol), Zn dust (4.45g, 68.16 mmol), and NH4Cl−THF (5:1, 42 mL) following the procedure as described in the preparation of 21. The usual work up and purification by column chromatography on silica gel (100−200 mesh) using petroleum ether−EtOAc (4:1) as eluent furnished the hydroxy allyl derivative ( $\beta$ isomer, 2.65 g, 67%) as a colorless liquid. To a cold (0 °C) suspension of NaH (60% in mineral oil, 0.343 g, 8.57 mmol) in dry DMF (10 mL) was added a solution of the above isomer (2.5 g, 7.14 mmol) in DMF (15 mL), and the mixture was stirred for 30 min at 0 °C. Benzyl bromide (1.10 mL, 9.28 mmol) in dry DMF (15 mL) containing TBAB (0.262 g, 0.81 mmol) was added to the reaction mixture, which was then stirred at room temperature for 6 h. The reaction was quenched with a saturated aqueous NH4Cl solution (25 mL), and the organic solvent was evaporated under vacuum to a residue, which was extracted with EtOAc  $(3 \times 30 \text{ mL})$ . The combined organic extract was dried  $(Na<sub>2</sub>SO<sub>4</sub>)$  and evaporated to a residue, which was purified by column chromatography on silica gel (60−120 mesh). Elution was made with a petroleum ether−EtOAc (19:1) mixture to give the benzyl protected olefin derivative (2.65 g, 84%) as a colorless gum. A solution of the above derivative  $(2.0 \text{ g})$  in DCM−H<sub>2</sub>O  $(20:1, 42 \text{ mL})$ was oxidized by DDQ (1.55 g, 6.82 mmol) following the procedure as described in the preparation of 4. The usual work up and purification by column chromatography on silica gel (100−200 mesh) using petroleum ether−EtOAc (5:1) produced 25 (1.2 g, 82%) as a colorless oil.  $[\alpha]_{\text{D}}^{25}$  – 5 (c 0.15, CHCl<sub>3</sub>); <sup>1</sup>H NMR (CDCl<sub>3,</sub> 300 MHz):  $\delta$  1.32 (s, 3H), 1.48 (s, 3H), 2.36−2.45 (m, 2H), 4.04−4.10 (m, 2H), 4.27  $(brs, 1H)$ , 4.32  $(brs, 1H)$ , 4.51  $(d, 1H, J = 3.3 Hz)$ , 4.66  $(d, 1H, J = 1.55)$ 11.4 Hz), 4.80 (d, 1H,  $J = 11.4$  Hz), 5.11 (d, 1H,  $J = 11.1$  Hz), 5.16 (d, 1H, J = 18.6 Hz), 5.76−5.90 (m, 1H), 5.98 (d, 1H, J = 3.6 Hz), 7.33− 7.43 (m, 5H); <sup>13</sup>C NMR (CDCl<sub>3,</sub> 75 MHz):  $\delta$  26.1 (CH<sub>3</sub>), 26.7  $(CH<sub>3</sub>)$ , 36.4 (CH<sub>2</sub>), 74.0 (CH<sub>2</sub>), 75.3 (CH), 78.0 (CH), 80.5 (CH), 85.2 (CH), 104.5 (CH), 111.5 (C), 118.0 (CH<sub>2</sub>), 128.0 (3  $\times$  CH), 128.4 (2  $\times$  CH), 133.5 (CH), 137.6 (C); HRMS (EI, magnetic sector, positive ion) calcd for  $C_{18}H_{24}O_5$ ,  $m/z$  320.1624, found 320.1619.

(3aR,5R,5aS,6aR,9aR,9bR)-1-Benzyl-5-(benzyloxy)-8,8 dimethyloctahydro[1,3]dioxolo[4″,5″:4′,5′]furo[3′,2′:1,5]cyclopenta[1,2-c]isoxazole (27). Oxidation of the hydroxyl group of 25, followed by INC reaction of the generated ketone, was done, according to the procedure as described in 6, using 25 (2.0 g, 6.25 mmol), DMP (3.98 g, 9.38 mmol), and dry DCM (30 mL). The usual work up afforded a crude residue. To a solution of this residue in toluene (50 mL) was added BnNHOH (1.15 g, 9.38 mmol), and the mixture was heated at reflux for 12 h. The usual work up and purification by column chromatography on silica gel (230−400 mesh) using petroleum ether−EtOAc (4:1) afforded 27 (1.75 g, 66%) as a colorless solid. mp 120−121 °C;  $[\alpha]_{D}^{25}$  + 14.0 (c 0.16, CHCl<sub>3</sub>); <sup>1</sup>H NMR (CDCl<sub>3</sub>, 300 MHz):  $\delta$  1.36 (s, 3H), 1.58 (s, 3H), 1.87 (dd, 1H, J  $= 7.2, 12.6$  Hz), 2.19 (q, 1H,  $J = 10.5$  Hz), 2.69 (apparent t, 1H,  $J =$ 7.2, 7.8 Hz), 3.55 (d, 1H, J = 8.7 Hz), 3.86 (d, 1H, J = 13.8 Hz), 3.96 (apparent t, 1H,  $J = 6.9$ , 8.1 Hz), 4.11 (apparent t, 1H,  $J = 6.6$ , 8.1 Hz), 4.39 (d, 1H,  $J = 14.1$  Hz), 4.61 (d, 1H,  $J = 11.7$  Hz), 4.67 (merged s, 1H), 4.69 (d, 1H, J = 11.7 Hz), 4.77 (s, 1H), 5.88 (s, 1H), 7.30−7.54 (m, 10H); 13C NMR (CDCl3, 75 MHz): δ 26.6 (CH3), 27.2 (CH3), 35.2 (CH<sub>2</sub>), 48.6 (CH), 57.7 (CH<sub>2</sub>), 72.2 (CH<sub>2</sub>), 73.1 (CH<sub>2</sub>), 79.3 (CH), 80.1 (CH), 81.5 (C), 85.3 (CH), 105.2 (CH), 113.6 (C), 127− 128.4 (10 × CH), 138.0 (C), 138.4 (C); HRMS (EI, magnetic sector, positive ion) calcd for  $C_{25}H_{29}NO_5$ ,  $m/z$  423.2046, found 423.2051.

(2S,3R,4R,5R)-2-(Allyloxy)-4-(benzyloxy)-5-((benzyloxy) **methyl)tetrahydrofuran-3-ol (29).** To a mixture of 28 (2.5 g, 6.76 mmol) and dry allyl alcohol (50 mL) was added tosic acid (0.20 g, 1.08 mmol), and the mixture was heated at reflux for 6 h. The reaction mixture was neutralized with a saturated NaHCO<sub>3</sub> solution (30 mL), and the solvent was removed in vacuo until a syrupy residue was obtained. The residue was extracted with DCM  $(3 \times 40 \text{ mL})$ , and the organic layer was washed with  $H_2O$  (50 mL), dried, and concentrated to give a mixture of crude product ( $\alpha$  and  $\beta$  anomer). The mixture was separated and purified by column chromatography on silica gel (230− 400 mesh) using petroleum ether−EtOAc (9:1) as eluent to furnish 29  $(1.75 \text{ g}, 70\%)$  as a colorless oil.  $[\alpha]_{D}^{25} - 87$  (c 0.12, CHCl<sub>3</sub>); <sup>1</sup>H NMR  $(CDCI<sub>3</sub>, 300 MHz): \delta 2.77$  (d, 1H, J = 7.5 Hz), 3.65 (dd, 1H, J = 6.6,

10.5 Hz), 3.73 (dd, 1H,  $J = 4.2, 10.5$  Hz), 4.03 (dd, 1H,  $J = 4.2, 5.7$ Hz), 4.10 (dd, 1H,  $J = 6.3$ , 12.9 Hz), 4.26 (dd, 1H,  $J = 4.5$ , 7.8 Hz), 4.33 (dd, 1H,  $J = 5.1$ , 12.9 Hz), 4.42(dd, 1H,  $J = 5.7$ , 10.5 Hz), 4.52 (d, 1H,  $J = 12.0$  Hz), 4.55 (d, 1H,  $J = 12.0$  Hz), 4.62 (d, 1H,  $J = 12.0$  Hz), 4.74 (d, 1H,  $J = 12.0$  Hz), 5.14 (d, 1H,  $J = 4.8$  Hz), 5.20 (d, 1H,  $J =$ 10.5 Hz), 5.28 (dd, 1H, J = 1.2, 17.1 Hz), 5.84−5.97 (m, 1H), 7.27− 7.38 (m, 10H); <sup>13</sup>C NMR (CDCl<sub>3</sub>, 75 MHz):  $\delta$  68.99 (CH<sub>2</sub>), 69.0  $(CH<sub>2</sub>)$ , 71.8  $(CH<sub>2</sub>)$ , 73.4  $(CH<sub>2</sub>)$ , 76.9  $(CH)$ , 77.4  $(CH)$ , 83.5  $(CH)$ , 99.9 (CH), 117.6 (CH<sub>2</sub>), 127.5−128.3 (10 × CH), 133.7 (CH), 137.9 (C), 138.1 (C); HRMS (ESI−QToF, positive ion) calcd for  $C_{22}H_{26}NaO_5$ ,  $m/z$  393.1678, found 393.1658.

(3aS,5aS,7R,8R,8aS)-1-Benzyl-8-(benzyloxy)-7-(benzyloxy) methylhexahydro-1H-furo[2′,3′:2,3]furo[3,4-c]isoxazole (30). Oxidation and INC reaction were carried out, according to the procedure as described in 6, using 29 (0.350 g, 0.95 mmol), DMP (0.606 g, 1.43 mmol), and dry DCM (15 mL). The usual work up furnished a crude ketone, which was dissolved in toluene (10 mL), and then BnNHOH (0.176 g, 1.43 mmol) was added to it. The mixture was heated at reflux for 12 h. The usual work up and purification by column chromatography on silica gel (230−400 mesh) using petroleum ether−EtOAc (3:17) as eluent gave 30 (0.355 g, 79%) as a colorless solid. mp 175−176 °C; [ $\alpha$ ] $_{\text{D}}^{25}$  – 10 ( $\epsilon$  0.16, CHCl<sub>3</sub>); <sup>1</sup>H NMR (CDCl<sub>3</sub>, 300 MHz): δ 2.91−2.96 (m, 1H), 3.70−3.72 (m, 2H), 3.85 (dd, 1H,  $J = 2.1$ , 8.7 Hz), 3.93 (dd, 1H,  $J = 3.0$ , 9.3 Hz), 4.08 (d, 1H, J = 14.1 Hz), 4.04−4.11 (merged dd, 1H), 4.20 (d, 1H, J = 3.0 Hz), 4.24 (apparent t, 1H,  $J = 7.2$ , 9.3 Hz), 4.35 (dd, 1H,  $J = 3.0$ , 6.3 Hz), 4.40 (d, 1H,  $J = 14.7$  Hz), 4.49 (d, 1H,  $J = 11.7$  Hz), 4.56 (d, 1H,  $J = 11.7$  Hz), 4.65 (d, 1H,  $J = 11.1$  Hz), 4.91 (d, 1H,  $J = 11.4$  Hz), 6.02 (s, 1H), 7.17-7.39 (m, 15H); <sup>13</sup>C NMR (CDCl<sub>3</sub>, 75 MHz):  $\delta$  53.6 (CH), 58.5 (CH<sub>2</sub>), 68.2 (CH<sub>2</sub>), 71.9 (CH<sub>2</sub>), 73.1 (CH<sub>2</sub>), 73.7 (CH<sub>2</sub>), 74.8 (CH<sub>2</sub>), 80.9 (CH), 83.9 (CH), 86.7 (C), 104.3 (CH), 127.0− 128.4 (15 × CH), 137.6 (C), 137.8 (C), 138.3 (C); HRMS (ESI− QToF, positive ion) calcd for  $C_{29}H_{31}NNaO_5$ ,  $m/z$  496.2100, found 496.2108.

(3aS,5aR,6R,8aS)-1-Benzyl-6-((benzyloxy)methyl)hexahydro-1H-furo[3′,4′:2,3]furo[3,4-c]isoxazole (32). Oxidation, nitrone formation, and in situ cyclization were executed, following the procedure as described in 6, using 31 (0.550 g, 2.10 mmol), DMP (1.34 g, 3.15 mmol), and dry DCM (20 mL). The usual work up afforded a ketone, which was dissolved in toluene (15 mL), and BnNHOH (0.390 g, 3.15 mmol) was added to it. The mixture was heated at reflux for 6 h. The usual work up and purification by column chromatography on silica gel (230−400 mesh) using petroleum ether−EtOAc (4:1) furnished 32 (0.475 g, 62%) as a colorless solid. mp 170−171 °C; [a]<sup>25</sup> − 37 (c 0.25, CHCl<sub>3</sub>); <sup>1</sup>H NMR (CDCl<sub>3</sub>, 300 MHz): δ 3.11−3.13 (m, 1H), 3.67−3.85 (m, 7H), 3.96−3.99 (m, 1H), 4.14−4.19 (m, 2H), 4.24 (d, 1H, J = 9.9 Hz), 4.54 (s, 1H), 4.56 (partially merged d, 1H,  $J = 12.0$  Hz), 4.64 (d, 1H,  $J = 12.0$  Hz), 7.28– 7.36 (m, 10H); 13C NMR (CDCl3, 75 MHz): δ 54.6 (CH), 55.9  $(CH<sub>2</sub>)$ , 68.2 (CH<sub>2</sub>), 70.6 (CH<sub>2</sub>), 72.0 (CH<sub>2</sub>), 73.6 (CH<sub>2</sub>), 74.6 (CH<sub>2</sub>), 82.7 (2 × CH), 87.1 (C), 127.4−128.4 (10 × CH), 137.0 (C), 138.0 (C); HRMS (EI, magnetic sector, positive ion) calcd for  $C_{22}H_{25}NO_4$ , m/z 367.1784, found 367.1779.

(3aR,7R,7aS)-6-(Allyloxy)-2,2-dimethyltetrahydro-3aH-[1,3] dioxolo[4,5-c]pyran-7-ol (34). A mixture of 33 (10 g, 66.7 mmol), allyl alcohol (150 mL), and  $p$ -TSA·H<sub>2</sub>O (1.27 mg, 6.67 mmol) was heated at reflux for 4 h under  $N<sub>2</sub>$ . The reaction mixture was filtered, and the filtrate was evaporated in vacuo to a residue, which was extracted with  $CHCl<sub>3</sub>$  (2 × 50 mL). The combined extract was washed with brine (50 mL), dried  $(Na_2SO_4)$ , and evaporated to a crude anomeric mixture of the allylated product (11.4 g), which was used without further purification for the next step. A solution of 2,2 dimethoxy propane (9.7 mL, 78.95 mmol) and  $p$ -TSA·H<sub>2</sub>O (1.5 g, 7.89 mmol) in dry acetone (150 mL) was added to the above crude anomeric mixture (10.0 g, 52.6 mmol), and the mixture was stirred under  $N_2$  for 12 h. Silver carbonate (3.26 g, 11.8 mmol) was added, and the mixture was stirred for another 50 min. The heterogeneous mixture was filtered, and the residue was washed with acetone (100 mL). The combined solvent was evaporated under vacuum to a crude mixture (anomers) of products, which were purified by column

chromatography on silica gel (230−400 mesh) using petroleum ether−EtOAc (4:1) as eluent to furnish (34, β-anomer, 4.6 g, 38%) as a colorless gum. <sup>1</sup>H NMR (CDCl<sub>3</sub>, 300 MHz):  $\delta$  1.39 (s, 3H), 1.60 (s, 3H), 2.19 (d, 1H, J = 8.7 Hz), 3.55−3.69 (m, 2H), 3.83 (dd, 1H, J = 3.9, 10.8 Hz), 3.97−4.11 (m, 2H), 4.26 (m, 1H), 4.42 (brs, 1H), 4.75  $(d, 1H, J = 3.3 Hz)$ , 5.21  $(d, 1H, J = 10.4 Hz)$ , 5.31  $(d, 1H, 17.4 Hz)$ , 5.87–5.94 (m, 1H); <sup>13</sup>C NMR (CDCl<sub>3</sub>, 75 MHz): 25.3 (CH<sub>3</sub>), 26.5 (CH<sub>3</sub>), 61.8 (CH<sub>2</sub>), 68.3 (CH), 69.0 (CH<sub>2</sub>), 72.7 (CH), 72.9 (CH), 99.4 (CH), 109.8 (C), 117.5 (CH<sub>2</sub>), 133.9 (CH); ESIMS, m/z: 253  $(M + Na)^+$ .

(3aR,5aR,7aR,10aS,10bR)-1-Benzyl-9,9-dimethyloctahydro- [1,3]dioxolo[4″,5″:4′,5′]pyrano[2′,3′:2,3]furo[3,4-c]isoxazole (35). Oxidation and subsequent INC reaction were performed, following the procedure as described in 6, using 34 (1.0 g, 4.35 mmol), DMP (2.03 g, 4.78 mmol), and dry DCM (100 mL). The usual work up produced a crude residue, which was treated with BnNHOH (803 mg, 6.53 mmol) in refluxing toluene (25 mL) for 12 h. The usual work up and purification by a column chromatography on silica gel (230−400 mesh) using petroleum ether−EtOAc (10:1) as eluent gave 35 (800 mg, 55%) as solid material. mp 164−165 °C;  $[\alpha]_D^{25}$  + 15 (c 0.11, CHCl<sub>3</sub>); <sup>1</sup>H NMR (CDCl<sub>3</sub>, 600 MHz):  $\delta$  1.41 (s, 3H), 1.51 (s, 3H), 2.88 (quint, 1H,  $J = 4.2$  Hz), 3.66 (dd, 1H,  $J = 3.6$ , 13.2 Hz), 3.76 (dd, 1H  $J = 3.6$ , 12.6 Hz), 3.78 (d, 1H,  $J = 9.0$  Hz), 3.85  $(dd, 1H, J = 4.2, 9.0 Hz$ ), 3.91 (dd, 1H, J = 4.8, 9.0 Hz), 4.07 (d, 1H, J  $= 13.8$  Hz), 4.33–4.36 (m, 2H), 4.44 (t, 1H, J = 9.0 Hz), 4.71 (d, 1H, J = 6.6 Hz), 5.56 (s, 1H), 7.26–7.43 (m, 5H); <sup>13</sup>C NMR (CDCl<sub>3</sub>, 75 MHz):  $\delta$  25.5 (CH<sub>3</sub>), 27.0 (CH<sub>3</sub>), 51.1 (CH), 57.0 (CH<sub>2</sub>), 64.1  $(CH_2)$ , 71.9 (CH<sub>2</sub>), 72.8 (CH), 74.3 (C), 75.2 (CH<sub>2</sub>), 76.2 (CH), 100.2 (CH), 110.0 (C), 127.1 (CH), 128.5 (2 × CH), 128.6 (2 × CH), 138.7 (C); HRMS (EI, magnetic sector, positive ion) calcd for  $C_{18}H_{23}NO_5$ ,  $m/z$  333.1576, found 333.1570.

(3aR,4aS,8S,9aR,9bR)-9a-Amino-2,2-dimethyloctahydro- [1,3]dioxolo[4′, 5′:4, 5]furo[2, 3-c]oxepin-8-ol (36). To a stirred solution of 6 (1.5 g, 4.5 mmol) in CH<sub>3</sub>CN-H<sub>2</sub>O (15:1, 16 mL) was added  $Mo(CO)_{6}$  (1.80 g, 6.8 mmol), and the mixture was heated at reflux under  $N_2$  for 12 h. The solvent was removed in vacuo, and the residue was dissolved in a DCM−MeOH mixture (15:1) and was passed through a bed of neutral alumina. The solvent was evaporated to give 36 (0.85 g, 77%) as a thick oil.  $[\alpha]_{\rm D}^{25}$  + 31 (c 0.10, CHCl<sub>3</sub>); <sup>1</sup>H NMR (CDCl<sub>3</sub>, 300 MHz): δ 1.35 (s, 3H), 1.55 (s, 3H), 1.95−2.06 (m, 2H), 2.18 (brs, 3H), 3.82 (s, 1H), 3.91 (d, 2H, J = 3.6 Hz), 4.05 (d, 1H, J = 3.6 Hz), 4.07 (d, 2H, J = 2.7 Hz), 4.18−4.20 (m, 1H), 5.87 (d, 1H,  $J = 3.9$  Hz); <sup>13</sup>C NMR (CDCl<sub>3</sub>, 150 MHz):  $\delta$  26.5 (CH<sub>3</sub>), 26.8  $(CH_3)$ , 34.1 (CH<sub>2</sub>), 64.3 (C), 67.7 (CH<sub>2</sub>), 71.3 (CH), 78.7 (CH<sub>2</sub>), 81.5 (CH), 86.1 (CH), 103.4 (CH), 112.0 (C); HRMS (EI, magnetic sector, positive ion) calcd for  $C_{11}H_{19}NO_5$ ,  $m/z$  245.1263, found 245.1260.

(3aR,4aR,8S,9aR,9bR)-tert-Butyl-9a-amino-8-hydroxy-2,2 dimethylhexahydro-3aH-[1,3]dioxolo[4′,5′:4,5]furo[2,3-c] azepine-6(9bH)-carboxylate (37). Cleavage of the N−O bond and debenzylation were carried out, following the procedure as described in 36, using 18 (1.2 g, 2.78 mmol),  $Mo(CO)_{6}$  (1.1 g, 4.17 mmol), and a CH3CN−H2O mixture (15:1, 16 mL). The usual work up provided 37 (0.80 g, 84%) as a yellow oil.  $[\alpha]_D^{25}$  + 35 (c 0.23, CHCl<sub>3</sub>); <sup>1</sup>H NMR  $(CDCl<sub>3</sub> + D<sub>2</sub>O<sub>3</sub> 300 MHz): \delta 1.32$  (s, 3H), 1.46 (s, 9H), 1.54 (s, 3H), 1.82 (apparent t, 2H,  $J = 16.0$  Hz), 3.28 (dd, 1H,  $J = 3.6$ , 13.8 Hz), 3.53−3.77 (m, 2H), 3.84−3.92 (m, 1H), 3.96 (s, 1H), 4.07−4.26 (m, 2H), 5.74 (s, 1H); <sup>13</sup>C NMR (CDCl<sub>3,</sub> 75 MHz): δ 26.3 (CH<sub>3</sub>), 26.4  $(CH_3)$ , 26.6 (CH<sub>3</sub>), 28.3 (3 × CH<sub>3</sub>), 35.7 (CH<sub>2</sub>), 35.9 (CH<sub>2</sub>), 47.4  $(CH<sub>2</sub>)$ , 48.7 (CH<sub>2</sub>), 55.0 (CH<sub>2</sub>), 56.5 (CH<sub>2</sub>), 64.4 (C), 64.9 (C), 69.6 (CH), 70.4 (CH), 77.2 (C), 79.8 (CH), 80.0 (CH), 86.6 (CH), 87.2 (CH), 103.1 (CH), 103.4 (CH), 111.6 (C), 111.8 (C), 155.7 (C), 155.9 (C); HRMS (ESI−QToF, positive ion) calcd for  $C_{16}H_{28}N_2NaO_6$ ,  $m/z$  367.1845, found 367.1827.

((3aR,4aS,5R,7R,7aR,7bR)-7a-Amino-5-(benzyloxy)-2,2 dimethylhexahydro-3aH-cyclo-penta[4,5]furo[2,3-d][1,3] **dioxol-7-yl)methanol (38).** A solution of  $27$  (1.5 g, 3.55 mmol) in CH<sub>3</sub>CN−H<sub>2</sub>O (15:1, 16 mL) was treated with Mo(CO)<sub>6</sub> (1.87 g, 7.1) mmol), according to the procedure as described in 36. The usual work up gave 38 (0.85 g, 71%) as a thick oil.  $[\alpha]_D^{25}$  + 77 (c 0.11, CHCl<sub>3</sub>); <sup>1</sup>H

NMR (CDCl<sub>3</sub>, 300 MHz): δ 1.35 (s, 3H), 1.56 (s, 3H), 1.76−1.85 (m, 1H), 1.99−2.10 (m, 2H), 2.32 (brs, 3H), 3.63 (dd, 1H, J = 9.0, 11.1 Hz), 3.73 (dd, 1H,  $J = 4.8$ , 11.4 Hz), 4.08 (dt, 1H,  $J = 3.3$ , 8.1 Hz), 4.18 (d, 1H,  $J = 3.3$  Hz), 4.41 (d, 1H,  $J = 3.6$  Hz), 4.55 (d, 1H,  $J = 12.0$ Hz), 4.62 (d, 1H, J = 12.0 Hz), 5.94 (d, 1H, J = 3.6 Hz), 7.28–7.37 (m, 5H); <sup>13</sup>C NMR (CDCl<sub>3,</sub> 75 MHz):  $\delta$  26.6 (CH<sub>3</sub>), 27.1 (CH<sub>3</sub>),  $31.1(CH<sub>2</sub>), 42.4 (CH), 62.8 (CH<sub>2</sub>), 69.7 (C), 71.8 (CH<sub>2</sub>), 78.5 (CH),$ 85.6 (CH), 86.9 (CH), 106.1 (CH), 112.5 (C), 127.7 (CH), 127.8 (2 × CH), 128.3 (2 × CH), 138.0 (C); HRMS (ESI−QToF, positive ion) calcd for  $C_{18}H_{25}NNaO_5$ ,  $m/z$  358.1630, found 358.1628.

(3aR,4aS,8S,9aS,9bR)-9a-(Benzylamino)-2,2-dimethyloctahydrothiepino[4′,3′:4,5]furo[2,3-d][1,3]dioxol-8-ol (39). Isoxazolidine ring cleavage of 11 (250 mg, 0.72 mmol) in a mixture of CH<sub>3</sub>CN−H<sub>2</sub>O (15:1, 8 mL) was done following the procedure as described in 36, using  $Mo(CO)_{6}$  (285 mg, 1.08 mmol). The usual work up afforded 39 (175 mg, 69%) as a yellow oil.  $[\alpha]_D^{25} + 40$  (c 0.37, CHCl<sub>3</sub>); <sup>1</sup>H NMR (CDCl<sub>3,</sub> 600 MHz):  $\delta$  1.40 (s, 3H), 1.56 (s, 3H), 2.01 (dd, 1H,  $J = 1.8$ , 15.6 Hz), 2.25 (dd, 1H,  $J = 4.8$ , 14.4 Hz), 2.82  $(dt, 2H, J = 5.4, 14.4 Hz), 3.01 (dd, 1H, J = 4.8, 14.4 Hz), 3.31 (dd,$ 1H,  $J = 3.6$ , 15.6 Hz), 3.96 (d, 1H,  $J = 11.4$  Hz), 4.08 (brs, 1H), 4.16 (d, 1H, J = 10.8 Hz), 4.38 (d, 1H, J = 3.0 Hz), 4.47–4.50 (m, 1H), 5.83 (d, 1H, J = 4.2 Hz), 7.28–7.37 (m, 5H), two Hs not discernible; <sup>13</sup>C NMR (CDCl<sub>3</sub>, 150 MHz):  $\delta$  26.6 (CH<sub>3</sub>), 26.9 (CH<sub>3</sub>), 31.1 (CH<sub>2</sub>), 33.7 (CH<sub>2</sub>), 42.2 (CH<sub>2</sub>), 48.0 (CH<sub>2</sub>), 70.3 (CH), 71.2 (C), 77.2 (CH), 84.8 (CH), 103.0 (CH), 112.2 (C), 128.4−128.9 (5 × CH), one (C) not discernible; HRMS (ESI−QToF, positive ion) calcd for  $C_{18}H_{26}NO_4S$ ,  $m/z$  352.1583, found 352.1612.

((2R,3R,3aS,4S,6aS)-3a-Acetamido-3-acetoxyhexahydrofuro- [2,3-b]furan-2,4-diyl)bis(methylene)diacetate (40). Pd/C (10%, 100 mg) was added to a solution of 30 (600 mg, 1.27 mmol) in MeOH (20 mL) and hydrogenated with  $H_2$  gas under 1 atmospheric pressure at room temperature for 12 h. The catalyst was filtered off, the solvent was evaporated, and the residue was used in the next step without further purification. The residue (210 mg, 1.02 mmol) was dissolved in pyridine  $(20 \text{ mL})$ , Ac<sub>2</sub>O  $(1.0 \text{ mL}, 10.2 \text{ mmol})$  and DMAP (pinch) were added to the solution, and the mixture was stirred at room temperature for 6 h. Pyridine was evaporated through azeotropic distillation with toluene under vacuum. The residue was extracted with CHCl<sub>3</sub> ( $3 \times 40$  mL), the combine extract was washed with brine ( $50$ mL), and dried  $(Na_2SO_4)$ , and the solvent was evaporated to a residue, which was purified by column chromatography on silica gel (230−400 mesh) using petroleum ether−EtOAc (2:3) as eluent to afford 40 (350 mg, 74% overall) as a thick oil.  $[\alpha]_{D}^{25} - 7$  (c 0.14, CHCl<sub>3</sub>); <sup>1</sup>H NMR  $(CDCl_3, 300 MHz)$ :  $\delta$  2.03 (s, 3H), 2.08 (s, 3H), 2.09 (s, 3H), 2.12 (s, 3H), 2.73 (quint, 1H, J = 7.2 Hz), 4.00−4.06 (m, 2H), 4.12−4.30 (m, 4H), 4.56 (td, 1H,  $J = 4.5$ , 7.8 Hz), 5.49 (d, 1H,  $J = 3.3$  Hz), 5.91 (s, 1H), 5.99 (s, 1H); <sup>13</sup>C NMR (CDCl<sub>3</sub>, 75 MHz): δ 20.4 (CH<sub>3</sub>), 20.7  $(CH_3)$ , 20.7 (CH<sub>3</sub>), 23.0 (CH<sub>3</sub>), 45.5 (CH), 61.7 (2  $\times$  CH<sub>2</sub>), 71.2  $(CH<sub>2</sub>)$ , 72.8 (C), 76.7 (CH), 78.0 (CH), 110.6 (CH), 170.0 (C), 170.6 (C), 170.96 (C), 170.99 (C); HRMS (ESI−QToF, positive ion) calcd for  $C_{16}H_{23}NNaO_9$ ,  $m/z$  396.1271, found 396.1257.

((3S,3aS,6R,6aR)-3a-Acetamidohexahydrofuro[3,4-b]furan-3,6-diyl)bis(methylene)diacetate (41). Isoxazolidine ring cleavage, followed by debenzylation and acetylation, was carried out, following the procedure as described in 40, using 32 (500 mg, 1.36 mmol) in MeOH (20 mL), Pd/C (10%, 80 mg), pyridine (20 mL), Ac<sub>2</sub>O (0.81 mL), and DMAP (pinch). The usual work up and purification by column chromatography on silica gel (230−400 mesh) using petroleum ether−EtOAc (1:1) as eluent afforded 41 (305 mg, 71% overall) as a thick oil.  $[\alpha]_{\text{D}}^{25} - 3$  (c 0.07, CHCl<sub>3</sub>); <sup>1</sup>H NMR (CDCl<sub>3</sub>, 300 MHz): δ 2.02 (s, 3H), 2.08 (s, 3H), 2.10 (s, 3H), 2.82−2.86 (m, 1H), 3.89−3.95 (m, 2H), 4.00−4.07 (m, 3H), 4.16−4.24 (m, 3H), 4.29 (d, 1H, J = 10.2 Hz), 4.41 (d, 1H, J = 3.6 Hz), 6.20 (brs, 1H); <sup>13</sup>C NMR (CDCl<sub>3</sub>, 150 MHz):  $\delta$  20.9 (2  $\times$  CH<sub>3</sub>), 23.2 (CH<sub>3</sub>), 47.4 (CH), 62.6 (CH<sub>2</sub>), 62.7 (CH<sub>2</sub>), 71.4 (C), 72.1 (CH<sub>2</sub>), 79.3 (CH<sub>2</sub>), 80.2 (CH), 87.9 (CH), 170.6 (C), 170.88 (C), 170.91 (C); HRMS (ESI− QToF, positive ion) calcd for  $C_{14}H_{21}NNaO_7$ ,  $m/z$  338.1216, found 338.1197.

((2R,4aS,6aR,7R,9aS)-2-Methyloctahydrofuro[3′,4′:2,3]furo- [3,4-d][1,3]oxazin-7-yl)methanol (43). Hydrogenolysis of 32 (700 <span id="page-10-0"></span>mg, 1.91 mmol) in dry MeOH (20 mL) was carried out over Pd/C (10%, 100 mg) following the procedure as described in 40. The usual work up afforded a residue 42 (250 mg, 1.32 mmol), which was dissolved in 40% aqueous CH<sub>3</sub>CHO (w/w %), and the mixture was stirred at room temperature for 12 h. The solvent was evaporated, and the residue was extracted with EtOAc (100 mL). The organic layer was washed with brine (50 mL), dried  $(Na_2SO_4)$  and evaporated to a residue, which was purified by column chromatography on silica gel (230−400 mesh) using petroleum ether−EtOAc (4:6) as eluent to provide 43 (235 mg, 57%) as a solid material. mp 142−143 °C;  $[\alpha]_{\text{D}}^{25}$  $+$  3 (c 0.07, CHCl<sub>3</sub>); <sup>1</sup>H NMR (CDCl<sub>3</sub>, 300 MHz):  $\delta$  1.31 (d, 3H, J = 5.7 Hz), 2.04−2.12 (m, 1H), 2.47 (d, 1H,  $J = 6.6$  Hz), 3.50 (d, 1H,  $J =$ 11.7 Hz), 3.57 (dd, 1H, J = 2.7, 9.3 Hz), 3.79−4.13 (m, 8H), 4.27− 4.36 (m, 1H), one H not discernible; <sup>13</sup>C NMR (CDCl<sub>3</sub>, 75 MHz):  $\delta$ 21.5 (CH<sub>3</sub>), 39.9 (CH), 61.6 (CH<sub>2</sub>), 68.0 (CH<sub>2</sub>), 71.0 (CH<sub>2</sub>), 71.3  $(C)$ , 81.1  $(CH_2)$ , 81.8  $(CH)$ , 83.1  $(CH)$ , 83.6  $(CH)$ ; HRMS (ESI– QToF, positive ion) calcd for  $C_{10}H_{17}NNaO_4$ ,  $m/z$  238.1055, found 238.1028.

Conversion of 43 to 41. To a solution of 43  $(210 \text{ mg}, 0.98)$ mmol) in pyridine (15 mL) were added  $Ac_2O$  (0.7 mL, 7.4 mmol) and DMAP (pinch), and the mixture was stirred at room temperature for 6 h. Pyridine was evaporated through azeotropic distillation with toluene in a rotary evaporator. The residue was extracted with CHCl<sub>3</sub> ( $3 \times 40$ ) mL), and the combined extract was washed with brine (50 mL), dried  $(Na<sub>2</sub>SO<sub>4</sub>)$ , and evaporated to a crude product, which was purified by column chromatography over silica gel (230−400 mesh) using petroleum ether−EtOAc (1:1) as eluent to afford 41 (280 mg, 91%) as a thick oil.

(2R,3R,3aR,5S,8aS)-3a-Acetamido-2-(2,4-dioxo-3,4-dihydropyrimidin-1(2H)-yl)octahydro-furo[2,3-c]oxepine-3,5-diyl) **diacetate (44).** Ac<sub>2</sub>O (2.20 mL, 23.2 mmol) was added to 36 (0.70 g, 2.9 mmol) dissolved in HOAc (30 mL). The mixture was cooled to 0 °C, TfOH (0.009 mL, 0.1 mmol) was added to it, and the mixture was stirred for 2 h at room temperature. The reaction was quenched with a cold saturated  $\mathrm{NaHCO}_{3}$  solution (10 mL), and the mixture was extracted with DCM  $(3 \times 25 \text{ mL})$ . The combined solvent was dried  $(Na<sub>2</sub>SO<sub>4</sub>)$ , and evaporated to a crude anomeric mixture of products, dried via co-evaporation with anhydrous  $CH_3CN$  (2  $\times$  15 mL). Uracil (0.682 g, 6.09 mmol) and N,O-bis(trimethylsilyl)acetamide (2.13 mL, 8.7 mmol) were added to a solution of the above anomeric mixture in CH<sub>3</sub>CN (20 mL), and the mixture was heated at reflux for 45 min until the suspension became a clear solution. The reaction mixture was cooled to 0 °C, and TMSOTf (0.79 mL, 4.35 mmol) was added to it dropwise and heated at 50 °C for 17 h. CH<sub>3</sub>CN was evaporated under reduced pressure to a residue, to which was added a saturated NH<sub>4</sub>Cl solution (10 mL). The mixture was extracted with DCM ( $2 \times 25$  mL). The combined extract was dried  $(Na_2SO_4)$ , evaporated, and the crude product was purified by column chromatography over silica gel (230− 400 mesh) using petroleum ether−EtOAc (2:3) as eluent to furnish 44  $(0.550 \text{ g}, 45\%)$  as a colorless foam.  $[\alpha]_D^{25}$  + 58 (c 0.23, CH<sub>3</sub>OH); <sup>1</sup>H NMR (CD3OD, 300 MHz): δ 2.02 (s, 3H), 2.06 (s, 3H), 2.16 (s, 3H), 2.94 (dd, 1H,  $J = 11.1$ , 14.4 Hz), 3.44 (dd, 1H,  $J = 10.5$ , 11.4 Hz), 3.87 (dd, 1H, J = 1.8, 14.4 Hz), 4.17−4.24 (m, 3H), 5.06−5.15 (m, 2H), 5.56 (d, 1H,  $J = 7.5$  Hz), 5.82 (d, 1H,  $J = 8.1$  Hz), 6.26 (d, 1H,  $J = 7.5$ Hz), 7.78 (d, 1H,  $J = 8.1$  Hz), two Hs were not discernible; <sup>13</sup>C NMR (CD<sub>3</sub>OD, 75 MHz):  $\delta$  20.5 (CH<sub>3</sub>), 20.9 (CH<sub>3</sub>), 23.5 (CH<sub>3</sub>), 37.1 (CH<sub>2</sub>), 63.1 (C), 69.2 (CH), 74.9 (CH<sub>2</sub>), 76.5 (CH), 76.7 (CH<sub>2</sub>), 85.4 (CH), 86.8 (CH), 104.1 (CH), 141.3 (CH), 152.7 (C), 165.9 (C), 171.5 (C), 171.7 (C), 174.0 (C); HRMS (ESI−QToF, positive ion) calcd for  $C_{18}H_{23}N_3NaO_9$ ,  $m/z$  448.1332, found 448.1306.

(2R,3R,3aR,4R,6R,6aS)-3a-Acetamido-4-(acetoxymethyl)-2- (2,4-dioxo-3,4-dihydro-pyrimidin-1(2H)-yl)hexahydro-2Hcyclopenta[b]furan-3,6-diyldiacetate (45). Nucleosidation on 38 was carried out, following the method as described in 44, using 38  $(0.70 \text{ g}, 2.1 \text{ mmol})$ , Ac<sub>2</sub>O (1.98 mL, 21.0 mmol), HOAc (25 mL), and TfOH (0.006 mL, 0.07 mmol) for peracetylation. The usual work up afforded a residue, which was treated with uracil (0.494 g, 4.41 mmol) and N,O-bis(trimethylsilyl)acetamide (1.54 mL, 6.3 mmol) in refluxing CH<sub>3</sub>CN (20 mL). The reaction mixture was cooled to 0  $^{\circ}$ C, and TMSOTf (0.57 mL, 3.15 mmol) was added to it dropwise and heated

at 50 °C for 17 h. The usual work up and purification by column chromatography over silica gel (230−400 mesh) using petroleum ether−EtOAc (3:7) as eluent furnished 45 (0.40 g, 41%) as a foam.  $[\alpha]_D^{25}$  + 79 (c 0.37, CH<sub>3</sub>OH); <sup>1</sup>H NMR (CD<sub>3</sub>OD<sub>,</sub> 300 MHz):  $\delta$  1.96 (s, 3H), 1.98 (s, 3H), 2.00 (s, 3H), 2.04 (s, 3H), 2.08−2.16 (partially merged m, 1H), 2.27−2.31 (m, 1H), 2.66 (m, 1H), 3.96 (dd, 1H, J = 6.6, 8.1 Hz), 4.07 (dd, 1H, J = 7.5, 11.1 Hz), 4.15 (d, 1H, J = 8.4 Hz), 5.14 (d, 1H, J = 5.7 Hz), 5.25–5.26 (m, 1H), 5.76 (d, 1H, J = 8.4 Hz), 5.88 (d, 1H, J = 8.4 Hz), 7.69 (d, 1H, J = 8.4 Hz); <sup>13</sup>C NMR (CD<sub>3</sub>OD) 150 MHz):  $\delta$  20.8 (CH<sub>3</sub>), 21.0 (CH<sub>3</sub>), 21.1 (CH<sub>3</sub>), 23.2 (CH<sub>3</sub>), 35.1(CH<sub>2</sub>), 43.3 (CH), 46.4 (CH), 65.6 (CH<sub>2</sub>), 70.3 (C), 72.9 (CH), 78.3 (CH), 85.7 (CH), 104.2 (CH), 141.4 (CH), 152.4 (C), 171.8 (C), 172.0 (C), 172.4 (C), 172.8 (C), 174.6 (C); HRMS (ESI−QToF, positive ion) calcd for  $C_{20}H_{25}N_3NaO_{10}$ ,  $m/z$  490.1438, found 490.1428.

(3aR,4aR,8S,9aR,9bR)-tert-Butyl-9a-acetamido-8-acetoxy-2,2-dimethylhexahydro-3aH-[1,3]dioxolo[4′,5′:4,5]furo[2,3-c] azepine-6(9bH)-carboxylate (46). To a solution of 37 (0.85 g, 2.5 mmol) in pyridine  $(20 \text{ mL})$  were added Ac<sub>2</sub>O  $(1.18 \text{ mL}, 12.5 \text{ mmol})$ and DMAP (a pinch). The mixture was stirred at room temperature for 12 h. Pyridine was evaporated by azeotropic distillation with toluene under vacuum. The residue was extracted with DCM  $(2 \times 30)$ mL), and the combined extract was washed with brine (30 mL), dried  $(Na<sub>2</sub>SO<sub>4</sub>)$ , and evaporated to a crude residue, which was purified by column chromatography on silica gel (100−200 mesh) using petroleum ether−EtOAc (3:1) as eluent to furnish 46 (0.80 g, 75%) as a crystalline material. mp 178−179 °C;  $[\alpha]_D^{25}$  + 53 (c 0.16, CHCl<sub>3</sub>);<br><sup>1</sup>H NMP (CDCL 300 MHz), δ 1 31 (c) 1 37 (c) 1 46 (c) 1 51 (c) <sup>1</sup>H NMR (CDCl<sub>3</sub>, 300 MHz):  $\delta$  1.31 (s), 1.37 (s), 1.46 (s), 1.51 (s), 1.53 (s), 1.78 (brt,  $J = 15$  Hz), 1.96 (s), 1.98 (s), 2.01 (s), 2.04 (s), 2.88 (brdt,  $J = 4.2, 14.1$  Hz), 3.11 (dd,  $J = 8.7, 13.2$  Hz), 3.40 (dd,  $J =$ 3.9,15.3 Hz), 3.83 (m), 4.08−4.18 (m), 4.30 (t, J = 5.7 Hz), 4.71 (d, J  $= 3.0$  Hz), 4.76 (d, J = 3.6 Hz), 4.80 (brs), 5.05 (brs), 5.77 (d, J = 3.3 Hz), 5.83 (m), 6.06 (s); <sup>13</sup>C NMR (CDCl<sub>3</sub>, 75 MHz):  $\delta$  21.1 (CH<sub>3</sub>), 21.3 (CH<sub>3</sub>), 23.6 (CH<sub>3</sub>), 26.5 (CH<sub>3</sub>), 28.2 (CH<sub>3</sub>), 28.3 (CH<sub>3</sub>), 32.1  $(CH<sub>2</sub>)$ , 35.8 (CH<sub>2</sub>), 47.8 (CH<sub>2</sub>), 49.2 (CH<sub>2</sub>), 51.7 (CH<sub>2</sub>), 53.2 (CH<sub>2</sub>), 62.5 (C), 64.2 (C), 68.5 (CH), 70.6 (CH), 79.6 (CH), 80.2 (C), 80.4 (C), 82.7 (CH), 84.3 (CH), 103.7 (CH), 111.9 (C), 155.6 (C), 155.8 (C), 170.0 (C); HRMS (EI, magnetic sector, positive ion) calcd for  $C_{20}H_{32}N_2O_8$ ,  $m/z$  428.2159, found 428.2151.

# ■ ASSOCIATED CONTENT

#### **S** Supporting Information

Copies of  ${}^{1}H$  and  ${}^{13}C$  NMR spectra of all the new compounds; ORTEP diagrams; and CIFs of 6, 23, 27, 30, 35, 43, and 46. This material is available free of charge via the Internet at http://pubs.acs.org.

## ■ [AUTHOR INF](http://pubs.acs.org)ORMATION

# Corresponding Author

\*E-mail: sbmandal@iicb.res.in (S.B.M.).

#### Notes

The auth[ors declare no comp](mailto:sbmandal@iicb.res.in)eting financial interest.

## ■ ACKNOWLEDGMENTS

We thank CSIR, Govt. of India, for providing a Senior Research Fellowship (S.N.D.) and NIPER Fellowship (A.C. and N.T.). We also thank Dr. R. Natarajan (Scientist) and Dr. P. R. Maulik (Emeritus Scientist) of the institute for their helpful discussion on single-crystal X-ray analyses of the compounds.

# ■ REFERENCES

(1) (a) Jones, R. C. F.; Martin, J. N. In Synthetic Applications of 1,3- Dipolar Cycloaddition Chemistry Toward Heterocycles and Natural Products; Padwa, A., Pearson, W. H., Eds.; John Wiley & Sons, Inc: New York, 2002; Chapter 1, pp 1−81. (b) Merino, P. Nitrones and its Cyclic Analogues. In Science of Synthesis; Bellus, D., Padwa, A., Eds.; George Thieme: Stuttgart, Germany, 2004; Vol. 27, pp 511−580.

<span id="page-11-0"></span>(c) Merino, P. Nitrones and Cyclic Analogues. An Update. In Science of Synthesis; Schaumann, E., Ed.; George Thieme: Stuttgart, Germany, 2011; Vol. 2010/4, pp 325−403. (d) Baggiolini, E. G.; Lee, H. L.; Pizzolato, G.; Uskokovic, M. R. J. Am. Chem. Soc. 1982, 104, 6460− 6462. (e) Tufariello, J. J.; Meckler, H.; Senaratne, K. P. A. Tetrahedron 1985, 41, 3447−3453. (f) Wovkulich, P. M.; Uskokovic, M. R. J. Am. Chem. Soc. 1981, 103, 3956−3958. (g) Wovkulich, P. M.; Uskokovic, M. R. Tetrahedron 1985, 41, 3455−3462. (h) Funk, R. L.; Bolton, G. L.; Daggett, J. U.; Hansen, M. M.; Horcher, L. H. M. Tetrahedron 1985, 41, 3479−3495. (i) Iida, H.; Kasahara, K.; Kibayashi, C. J. Am. Chem. Soc. 1986, 108, 4687−4648. (j) Kong, K.; Enquist, J. A., Jr.; McCallum, M. E.; Smith, G. M.; Matsumaru, T.; Menhaji-Klotz, E.; Wood, J. L. J. Am. Chem. Soc. 2013, 135, 10890−10893. (k) Yang, D.; Micalizio, G. C. J. Am. Chem. Soc. 2012, 134, 15237−15240. (l) Krenske, E. H.; Agopcan, S.; Aviyente, V.; Houk, K. N.; Johnson, B. A.; Holmes, A. B. J. Am. Chem. Soc. 2012, 134, 12010−12015. (m) Davis, F. A.; Gaddiraju, N. V.; Theddu, N.; Hummel, J. R.; Kondaveeti, S. K.; Zdilla, M. J. J. Org. Chem. 2012, 77, 2345−2359. (n) Zhang, X.; Cividino, P.; Poisson, J.−F.; Shpak-Kraievskyi, P.; Lawent, M. Y.; Martel, A.; Dujardin, G.; Py, S. Org. Lett. 2014, 16, 1936−1939. (o) Nguyen, T. B.; Beauseigneur, A.; Martel, A.; Dhal, R.; Laurent, M.; Dujardin, G. J. Org. Chem. 2010, 75, 611−620.

(2) Ernst, B., Hart, G. W., Sinaÿ, P., Eds. Carbohydrates in Chemistry and Biology; Wiley-VCH: Weiheim, Germany, 2000; Vol. 1.

(3) (a) Tamura, O.; Mita, N.; Kusaka, N.; Suzuki, H.; Sakamoto, M. Tetrahedron Lett. 1997, 38, 429−432. (b) Faltin, F.; Fehring, V.; Kadyrov, R.; Arrieta, A.; Schareina, T.; Selke, R.; Miethchen, R. Synthesis 2001, 638−646. (c) Totani, K.; Takao, K.; Tadano, K. Synlett 2004, 2066−2080.

(4) (a) Shing, T. K. M.; Elsley, D. A.; Gillhouley, J. G. J. Chem. Soc., Chem. Commun. 1989, 1280−1282. (b) Ferrier, R. J.; Furneaux, R. H.; Prasit, P.; Tylor, P. C.; Brown, K. L.; Gainsford, G. J.; Diehl, J. W. J. Chem. Soc., Perkin Trans I 1983, 1621−1628.

(5) (a) Majumder, S.; Bhattacharjya, A.; Patra, A. Tetrahedron Lett. 1997, 38, 8581−8584. (b) Koumbis, A. E.; Gallos, J. K. Curr. Org. Chem. 2003, 7, 585–628. (c) Fišera, L. In Heterocycles from Carbohydrate Precursors; Sayed, E., Ashry, H. E., Eds.; Springer-Verlag: Berlin, 2007; Chapter 8, pp 287−323.

(6) (a) Patra, R.; Bar, N. C.; Roy, A.; Achari, B.; Ghoshal, N.; Mandal, S. B. Tetrahedron 1996, 52, 11265−11272. (b) Chakraborty, C.; Vyavahare, V. P.; Dhavale, D. D. Tetrahedron 2007, 63, 11984− 11990.

(7) (a) Sahabuddin, Sk.; Roy, A.; Drew, M. G. B.; Roy, B. G.; Achari, B.; Mandal, S. B. J. Org. Chem. 2006, 71, 5980−5992. (b) Singha, K.; Roy, A.; Dutta, P. K.; Tripathy, S.; Sahabuddin, Sk.; Achari, B.; Mandal, S. B. J. Org. Chem. 2004, 69, 6507−6011. (c) Roy, A.; Achari, B.; Mandal, S. B. Tetrahedron Lett. 2006, 47, 3875−3879.

(8) (a) Bhattacharjya, A.; Chattopadhyay, P.; McPhail, A. T.; McPhail, D. R. J. Chem. Soc., Chem. Commun. 1990, 1508−1509; 1991, 136 (Corrigendum). (b) Shing, T. K. M.; Zhong, Y.−L.; Mak, T. C. W.; Wang, R.; Xue, F. J. Org. Chem. 1998, 63, 414−415. (c) Bhattacharjee, A.; Datta, S.; Chattopadhyay, P.; Ghosal, N.; Kundu, A. P.; Pal, A.; Mukhopadhyay, R.; Chowdhury, S.; Bhattacharjya, A.; Patra, A. Tetrahedron 2003, 59, 4623−4639 and references cited therein.

(9) (a) Brandi, A.; Cardona, F.; Cicchi, S.; Cordero, F. M.; Goti, A. Chem.Eur. J. 2009, 15, 7808−7821. (b) Puchalska, E. M.; Rowicki, T.; Sas, W.; Stawińska, K. M. Tetrahedron 2013, 69, 9826−9831. (c) Shing, T. K. M.; Wong, W. F.; Ikeno, T.; Yamada, T. Org. Lett. 2007, 9, 207−209.

(10) (a) Dondoni, A.; Junquera, F.; Merchán, F. L.; Merino, P.; Scherrmann, M.-C.; Tejero, T. J. Org. Chem. 1997, 62, 5484−5496. (b) Dhavale, D. D.; Jachak, S. M.; Karche, N. P.; Trombini, C. Tetrahedron 2004, 60, 3009−3016. (c) Li, X.; Takahashi, H.; Ohtake, H.; Ikegami, S. Heterocycles 2003, 59, 547−571.

(11) (a) Piperno, A.; Giofrè, S. V.; Iannazzo, D.; Romeo, R.; Romeo, G.; Chiacchio, U.; Rescifina, A.; Piotrowska, D. G. J. Org. Chem. 2010, 75, 2798−2805. (b) Rong, J.; Roselt, P.; Plavec, J.; Chattopadhyaya, J. Tetrahedron 1994, 50, 4921−4936. (c) Mukherjee, S.; Mandal, S. B.;

Bhattacharjya, A. RSC Adv. 2012, 2, 8969−8978. (d) Ghosh, R.; Maity, J. K.; Achari, B.; Mandal, S. B. J. Org. Chem. 2010, 75, 2419−2422. (e) Tripathi, S.; Roy, B. G.; Drew, M. G. B.; Achari, B.; Mandal, S. B. J. Org. Chem. 2007, 72, 7427−7430. (f) Bar, N. C.; Roy, A.; Achari, B.; Mandal, S. B. J. Org. Chem. 1997, 62, 8948−8951.

(12) (a) Siriwardena, A.; Sonawane, D. P.; Bande, O. P.; Markad, P. R.; Yonekawa, S.; Tropak, M. B.; Ghosh, S.; Chopade, B. A.; Mahuran, D. J.; Dhavale, D. D. J. Org. Chem. 2014, 79, 4398-4404 and the references cited therein.

(13) (a) Liautard, V.; Desvergnes, V.; Martin, O. R. Tetrahedron: Asymmetry 2008, 19, 1999−2002. (b) Gao, Z.-X.; Wang, M.; Wang, S.; Yao, Z.-J. Org. Lett. 2009, 11, 3678−3681.

(14) Farr, R. A.; Peet, N. P.; Fang, N. S. Tetrahedron Lett. 1990, 31, 7109−7112.

(15) Ferrier, R. J.; Prasit, P. J. Chem. Soc., Chem. Commun. 1981, 983−985.

(16) Torrente, S.; Noya, B.; Paredes, M. D.; Alonso, R. J. Org. Chem. 1997, 62, 6710−6711.

(17) (a) Sharma, G. V. M; Reddy, K. R.; Sankar, A. R.; Kunwar, A. C. Tetrahedron Lett. 2001, 42, 8893−8896. (b) Shing, T. K. M.; Zhong, Y.-L. Tetrahedron 2001, 57, 1573−1579. (c) Collins, P. M.; Ashwood, M. S.; Eder, H.; Wright, S. H. B.; Kennedy, D. J. Tetrahedron Lett. 1990, 31, 2055−2058. (d) Bernet, V. B.; Vasella, A. Helv. Chim. Acta 1979, 62, 1990−2016 and references cited therein. (e) Silva, A. M. G.; Tome, A. C.; Neves, M.G. P. M. S.; Silva, A. M. S.; Cavaleiro, J. A. S.; ́ Perrone, D.; Dondoni, A. Tetrahedron Lett. 2002, 43, 603−605.

(18) (a) Torrente, S.; Noya, B.; Branchadell, V.; Alonso, R. J. Org. Chem. 2003, 68, 4772−4783. (b) Osborn, H. M. I.; Gemmel, N.; Harwood, L. M. J. Chem. Soc., Perkin Trans. 1 2002, 22, 2419−2438. (c) Tronchet, J. M. J.; Mihaly, M. E. Helv. Chim. Acta 1972, 55, 1266− 1271.

(19) (a) Ma, X.; Tang, Q.; Ke, J.; Zhang, J.; Wang, C.; Wang, H.; Li, Y.; Shao, H. Chem. Commun. 2013, 49, 7085−7087. (b) Gowrisankar, S.; Lee, K. Y.; Kim, J. N. Bull. Korean Chem. Soc. 2006, 27, 929−932. (c) Sridhar, P. R.; Reddy, G. M.; Seshadri, K. Eur. J. Org. Chem. 2012, 6228−6235.

(20) Asai, T.; Morita, S.; Shirata, N.; Yaniguchi, T.; Monde, K.; Sakurai, H.; Ozeki, T.; Oshima, Y. Org. Lett. 2002, 14, 5456−5459.

(21) (a) Harrar, K.; Reiser, O. Chem. Commun. 2012, 48, 3457− 3459. (b) Weiser, R.; Yue, W.; Reiser, O. Org. Lett. 2005, 7, 5353− 5356.

(22) Wang, G. K.; Lin, B. B.; Rao, R.; Zhu, K.; Quin, X. Y.; Xie, G. Y.; Quin, M. J. Nat. Prod. Res. 2013, 27, 1348−1352.

(23) Knapp, S.; Gore, V. K. Org. Lett. 2000, 2, 1391−1393.

(24) Ghosh, A. K.; Chapsal, B. D.; Baldridge, A.; Steffey, M. P.; Walters, D. E.; Koh, Y.; Amano, M.; Mitsuya, H. J. Med. Chem. 2011, 54, 622−634.

(25) Basu, S.; Ellinger, B.; Rizzo, S.; Deraeve, C.; Schü rmann, M.; Preut, H.; Arndt, H.-D.; Waldmann, H. Proc. Natl. Acad. Sci. U.S.A. 2011, 108, 6805−6810.

(26) Nicolaou, K. C.; Claremon, D. A.; Barnette, W. E. J. Am. Chem. Soc. 1980, 102, 6611−6612.

(27) (a) Debus, O. M.; Kurlmann, G. Epilepsia 2004, 45, 103−108. (b) Sall, K. N.; Greff, L. J.; Johnson-Pratt, L. R.; Delucca, P. T.; Polis, A. B.; Kolodny, A. H.; Fletcher, C. A.; Cassel, D. A.; Boyle, D. R.; Skobieranda, F. Opthalmology 2003, 110, 615−624. (c) Ilardi, E. A.; Vitaku, E.; Njardarson, J. T. J. Med. Chem. 2014, 57, 2832−2842. (d) Castanheiro, T.; Donnard, M.; Gulea, M.; Suffert, J. Org. Lett. 2014, 16, 3060−3063.

(28) (a) McReynolds, M. D.; Dougherty, J. M.; Hanson, P. R. Chem. Rev. 2004, 104, 2239−2258 and references cited therein. (b) Oppolzer, W. Pure Appl. Chem. 1990, 62, 1241−1250.

(29) (a) Norbeck, D. W.; Spanton, S.; Broder, S.; Mitsuya, H. Tetrahedron Lett. 1989, 30, 6263−6266. (b) Chu, C. K.; Beach, J. W.; Jeong, L. S.; Choi, B. G.; Comer, F. I.; Alves, A. J.; Schinazi, R. F. J. Org. Chem. 1991, 56, 6503−6505. (c) Merino, P. In Chemical Synthesis of Nucleoside Analogues; Wiley: Hoboken, NJ, 2013.

(30) Yadav, J. S.; Reddy, B. V. S. Carbohydr. Res. 2000, 329, 885−888.

- <span id="page-12-0"></span>(31) Oikawa, Y.; Yoshioka, T.; Yonemitsu, O. Tetrahedron Lett. 1982, 23, 885−888.
- (32) Dess, D. B.; Martin, J. C. J. Am. Chem. Soc. 1991, 113, 7277− 7287.
- (33) Garegg, P. J.; Samuelsson, B. J. Chem. Soc., Perkin Trans. 1 1980, 2866−2869.
- (34) Omura, K.; Sharma, A. K.; Swern, D. J. Org. Chem. 1976, 41, 957−962.
- (35) Staudinger, H.; Meyer, J. Helv. Chim. Acta 1919, 2, 635−646. (36) Jones, J. K. N.; Thompson, J. L. Can. J. Chem. 1957, 35, 955− 959.
- (37) Paquette, L. A.; Zeng, Q.; Tusi, H.-C.; Johnston, J. N. J. Org. Chem. 1998, 63, 8491−8509.
- (38) Dam, J. H.; Fristrup, P.; Madsen, R. J. Org. Chem. 2008, 73, 3228−3235.
- (39) Wolfrom, M. L.; Hanessian, S. J. Org. Chem. 1962, 27, 1800− 1804.
- (40) Yoshimura, Y.; Kitano, K.; Satoh, H.; Watanabe, M.; Miura, S.; Sakata, S.; Sasaki, T.; Matsuda, A. J. Org. Chem. 1996, 61, 822−823.
- (41) Mukherjee, S.; Roy, B. G.; Das, S. N.; Mandal, S. B. Tetrahedron Lett. 2012, 53, 4929−4932.
- (42) Cicchi, S.; Goti, A.; Brandi, A.; Guarna, A.; De Sarlo, F. Tetrahedron Lett. 1990, 31, 3351−3354.
- (43) (a) Vorbrü ggen, H.; Höfle, G. Chem. Ber. 1981, 114, 1256− 1268. (b) Wang, Z. In Comprehensive Organic Name Reactions and Reagents; Wiley: Hoboken, NJ, 2010.